# Anthracyclinones. Part 5. ${ }^{1}$ Synthesis of Some Anthracyclinones and 4Hydroxyanthracyclinones containing a Tertiary Methyl Carbinol Function in Ring A from D-Glucose Precursors 

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#### Abstract

Reaction of 3-C-Methyl-1,2-O-isopropylidene- $\alpha$-D-ribo-pentodialdo-1,4-furanose (8a) with leucoquinizarin ( $2,3,4 \mathrm{a}, 9 \mathrm{a}$-tetrahydroanthracene-1,4,9,10-tetraone) (1a) in alkaline solution followed by aerial oxidation gave mainly (5S)-3-C-methyl 1,2-O-isopropylidene-5-(quinizarin-2-yl)- $\alpha$-D-ribofuranose (9a), acid hydrolysis of which gave the quinizarinylpyranose (14a). Similarly 3-O-benzyl-3-C-methyl-1,2- $O$-isopropylidene- $\alpha$-D-ribopentodialdo-1,4-furanose ( $\mathbf{8 b}$ ) and leucoquinizarin gave the (5S)-1,4-furanose derivative (9c) from which the ( $5 S$ )-O-benzyl quinizarinyl pyranose (14b) was obtained. In contrast, the $O$-benzylfuranose (8b) with leucoquinizarin and DBU gave a mixture of (5R) and (5S)3-O-benzyl-3-C-methyl-1,2-O-isopropylidene-5-(quinizarin-2-yl)- $\alpha$-D-ribofuranose (9d) and (9c) respectively. The (5R)-derivative produced the ( $5 R$ ) quinizarinylpyranose (15a) from which the corresponding (10R)-anthracyclinone (19c) was obtained. Similarly the (10S)-anthracyclinones (19a) and (19b) respectively were prepared from (9a) and (9c) respectively, and the latter was debenzylated with boron trichloride to produce (19a). In a similar manner the $O$-benzyl aldehydo sugar (8b) with 5 -hydroxyleucoquinizarin (1b) in DMF with DBN gave after aerial oxidation the (5R), (5S) and related 5-deoxy hydroxyglycitylanthraquinones (9e), (9f), and (9g) respectively. Each of these was converted into the corresponding (10R), (10S), and (10R) 7-deoxy-4-hydroxyanthracyclinones (19i), (19j), and (19h) by the same general series of reactions outlined above. Structures of the compounds were confirmed by UV, mass, IR, CD, and ${ }^{1} \mathrm{H}$ NMR spectroscopy.


In recent publications, ${ }^{1-5}$ we have described the synthesis of several fully substituted (in ring a) anthracyclinones in a sequence of reaction steps which uses at the first step a novel modification of the Marschalk reaction. ${ }^{6}$ This involves the condensation of an aldehydo-carbohydrate (which functions both as an aldehyde and a chiral template source of the ultimate ring a of the anthracyclinone and to some extent as a chiral controller of the newly introduced chiral centres) with leucoquinizarin (1a) $\dagger$ in aqueous alkaline solution or alternatively with DBU or DBN in dry DMF. Hydroxyglycitylor glycityl-quinizarins are first formed in good yield and readily converted by a simple series of reactions into anthracyclinones.

Included amongst the relatively few naturally occurring anthracyclinones or anthracyclines in which ring a is fully substituted are aranciamycin (2) and steffimycin (3). ${ }^{7-11}$ These compounds are also characterised by the presence of a tertiary methyl carbinol group in ring a. The same carbinol function is also present in the related antibiotic nogalamycin (4). ${ }^{12} \mathrm{We}$ have accordingly been especially interested to extend our new methods to include syntheses of anthracyclinones containing a tertiary methyl carbinol group in ring A. In addition, many of the naturally occurring anthracyclines contain additional hydroxy or alkoxy groups in ring D. Thus nogalamycin (4) contains hydroxy functions at the 1,4 -positions.

The carbohydrate derivative used in our experiments, namely 3 -C-methyl-1,2:5,6-di- $O$-isopropylidene- $\alpha$-d-allofuranose (5a) is readily obtained by reaction of the ketone (6) ${ }^{14}$ with methylmagnesium iodide. The methylcarbinol (5a) is readily converted into the diol (7a) with $75 \%$ aqueous acetic acid at room temperature overnight, and the diol with sodium metaperiodate produced the aldehyde (8a) in excellent yield. In
preliminary experiments the aldehyde (8a) with leucoquinizarin in methanol and tetrahydrofuran at $-10^{\circ} \mathrm{C}$ under nitrogen with aqueous sodium hydroxide over 1 h produced, after aerial oxidation and acidification of the solution, a good yield of (5S)-3-C-methyl-1,2-O-isopropylidene-5-(quinizarin-2-yl)- $\alpha$-D-ribofuranose (9a) which was obtained crystalline after a single chromatographic purification on silica gel. Confirmation for the assignment of structure (9a) to the product come from the latter's acetylation to produce a tetra- $O$-acetate (10a), its oxidation with pyridinium chlorochromate to the 5 -oxo derivative (11), elemental analysis, by its IR [e.g. $v_{\text {max }} 1380 \mathrm{~cm}^{-1}$ $\left(\mathrm{CMe}_{2}\right)$ ], mass [e.g. m/z $442\left(M^{+}\right), 427(M-15)$, and 269 (characteristic breakdown fragment $\mathrm{Q}-\mathrm{C} \stackrel{+}{\mathrm{H}}: \mathrm{OH} ; \mathrm{Q}=$ quin-izarin-2-yl)] and ${ }^{1} \mathrm{H}$ NMR [e.g. absence of signal for $2^{\prime}-\mathrm{H}$, signals at $\delta 2.90(\mathrm{~d}, 5-\mathrm{OH}), 5.3(\mathrm{dd}, 5-\mathrm{H})$, and $7.5\left(\mathrm{~s}, 3^{\prime}-\mathrm{H}\right)$, and full assignment of other protons)] spectra and by its subsequent reactions. The assignment of the ( $5 S$ )-stereochemistry was confirmed by comparison of the CD spectra of the compound with other related compounds of known configuration (Figure), e.g. (12) and (13) whose structures have been confirmed earlier by X-ray crystal studies. ${ }^{15}$ In this preliminary reaction there was little evidence for the formation of related ( $5 R$ ) and 5deoxy derivatives in contrast to results obtained during the preparation of the ethynyl derivative (9b). ${ }^{2}$ When (9a) was

[^0]
(1)
a; $R=H$ b; $\mathrm{R}=\mathrm{OH}$

(2) $\mathrm{R}^{1}=\mathrm{H}, \mathrm{R}^{2}=\mathrm{OH}$
(3) $R^{1}=O M e, R^{2}=H$

(4)

(5)
a; $R=H$
b; $R=B n$

(7)
a; $R=H$
b; $R=B n$

(6)

(6)
a; $R=H$
b; $R=B n$

(9)
a; $R^{1}=O H, R^{2}=R^{3}=R^{5}=H, R^{4}=M e$
b; $R^{1}=O H, R^{2}=R^{3}=R^{5}=H, R^{4}=C \equiv C H$
c; $R^{1}=O H, R^{2}=R^{5}=H, R^{3}=B n, R^{4}=M e$
d; $R^{1}=R^{5}=H, R^{2}=O H, R^{3}=B n, R^{4}=M e$
e; $R^{1}=H, R^{2}=R^{5}=O H, R^{3}=B n, R^{4}=M e$
$f ; R^{1}=R^{5}=O H, R^{2}=H, R^{3}=B n, R^{4}=M e$
g; $R^{1}=R^{2}=H, R^{3}=B n, R^{4}=M e, R^{5}=O H$
heated with $70 \%$ aqueous acetic acid for 3 h the isopropylidene group in (9a) was readily removed to

(11)

(12)
(13)
produce the pyranose derivative (14a); this was obtained by isomerisation of the first formed furanose derivative. The pyranose structure (14a) was confirmed by elemental analysis and mass and ${ }^{1} \mathrm{H}$ NMR spectroscopy. Thus, in the furanose derivative ( 9 a ) the benzylic hydrogen signal of $5-\mathrm{H}$ is a doublet ( $\delta 5.3$ ) coupled to the $5-\mathrm{OH}$ proton ( $J 3 \mathrm{~Hz}$ ); in the pyranose (14a) the corresponding $5-\mathrm{H}$ proton signal is only coupled to $4-\mathrm{H}$.

In the hope that periodate oxidation of (14a) might occur exclusively at the terminal diol the latter compound was treated with 1 equiv. of sodium metaperiodate in aqueous acetoneacetic acid at $0^{\circ} \mathrm{C}$ for 2 h . Under these conditions the starting material was completely converted into a major product, presumably the $O$-formyl derivative (16) or the isomeric (17a) [m/z $\left.400\left(M^{+}\right)\right]$although other compounds were also produced. The labile product was immediately treated with sodium dithionite in $2 \%$ aqueous sodium hydroxide to give a compound which was not isolated but was identical (TLC in different solvents) with the anthracyclinone (19a) prepared by an alternative route (see later). The tetra- $O$-acetyl derivative (10a) was also readily deblocked when heated with aqueous acetic acid to produce the diol (20) in which the acetyl groups had been retained. Structure (20) was confirmed for the product by elemental analysis and mass and ${ }^{1} \mathrm{H}$ NMR spectroscopy. Reaction of the diol (20) with periodate gave a compound in good yield to which the aldehydoerythrose structure (21) may be assigned and this was confirmed by mass [e.g. $\left.m / z 568\left(M^{+}\right)\right]$ and ${ }^{1} \mathrm{H}$ NMR spectroscopy. Thus 4 acetyl and one CMe groups are readily assigned, signals at 8.2 (s, OCHO) and 9.43 (s, CCHO) and full assignment of other protons. When the aldehyde (21) was treated with aqueous alkaline sodium dithionite in a nitrogen atmosphere a compound was isolated in small amount which was identical on TLC examination with the

anthracyclinone (19a) prepared as above and by an alternative route (see later).

To simplify isolation of the various reaction products and to isolate the diol system in (14a) we prepared by an improved method the $O$-benzyl allofuranose derivative (5b) ${ }^{16}$ as a crystalline solid by benzylation of the methylcarbinol (5a) with sodium hydride and benzyl chloride in dimethylsulphoxide (DMSO). This was readily converted into the aldehyde (8b) by treatment with aqueous acetic acid followed by sodium metaperiodate. Reaction of the aldehyde with leucoquinizarin in a mixture of tetrahydrofuran (THF), methanol, and aqueous sodium hydroxide under nitrogen for about 2 h gave after aerial oxidation and acidification, a precipitate of the (5S)hydroxyglycitylquinizarin (9c) which was readily purified by silica gel chromatography and recrystallised from ethanol as red needles ( $37 \%$ yield, m.p. $189^{\circ} \mathrm{C}$ ). Only one major isomer was obtained in this reaction namely the ( $5 S$ ) derivative (9c) analogous to the compound obtained using the unprotected aldehyde. The structure assigned to the $O$-benzyl derivative was confirmed by formation of a tri- $O$-acetate (10b), with acetic anhydride in pyridine, elemental analysis, mass, e.g. $m / z$ $532\left(M^{+}\right), 514\left(M-\mathrm{H}_{2} \mathrm{O}\right), 269(\mathrm{Q}: \stackrel{+}{\mathrm{C}} \mathrm{HOH} ; \mathrm{Q}=$ quinizarin-2-yl) and $91\left(100 \%, \mathrm{C}_{7} \mathrm{H}_{7}{ }^{+}\right)$and ${ }^{1} \mathrm{H}$ NMR [e.g. absence of a signal for $2^{\prime}-\mathrm{H}, 2.92\left(\mathrm{br}, 5-\mathrm{OH}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.0\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right)$, and $7.44\left(3^{\prime}-\mathrm{H}\right)$ ] spectra. Additionally the CD spectrum of the compound confirmed the ( $5 S$ ) configuration (Figure).

The ( $5 S$ ) derivative (9c) was readily and quantitatively deblocked when heated with $70 \%$ aqueous acetic acid to produce a single product which readily crystallised from ethanol or toluene as orange-red plates, m.p. $240^{\circ} \mathrm{C}$. The pyranose structure (14b) assigned to this compound was confirmed by elemental analysis and mass [e.g. $m / z 492\left(M^{+}\right), 474(M-$ $\left.\mathrm{H}_{2} \mathrm{O}\right)$ ] and ${ }^{1} \mathrm{H}$ NMR [e.g. $\delta 5.45(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H})$ and $4-\mathrm{H}$ coupled only to $\left.4-\mathrm{OH}, J_{4.4-\mathrm{OH}} 6.5 \mathrm{~Hz}\right]$ spectroscopy. Reaction of the

(18)
a; $\mathrm{R}=\mathrm{H}$
b; $\mathrm{R}=\mathrm{OH}$

a; $R^{1}=R^{4}=O H, R^{2}=R^{7}=H, R^{3}=M e, R^{5}, R^{6}=H, O H$ b; $R^{1}=O H, R^{2}=R^{7}=H, R^{3}=M e, R^{4}=O B n, R^{5}, R^{6}=H, O H$ c; $R^{1}=R^{7}=H, R^{2}=O H, R^{3}=M e, R^{4}=O B n, R^{5}, R^{6}=H, O H$ d; $R^{1}=O H, R^{2}=R^{7}=H, R^{3}=C \equiv C H, R^{4}=O B n, R^{5}, R^{6}=H, O H$ e; $R^{1}=R^{4}=O H, R^{2}=R^{5}=R^{6}=R^{7}=H, R^{3}=C \equiv C H$ $f ; R^{1}=O H, R^{2}=R^{5}=R^{6}=R^{7}=H, R^{3}=C \equiv C H, R^{4}=O B n$ g; $\mathrm{R}^{1}=\mathrm{OH}, \mathrm{R}^{2}=\mathrm{R}^{4}=\mathrm{R}^{7}=\mathrm{H}, \mathrm{R}^{3}=\mathrm{p}-\mathrm{ClC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{O}, \mathrm{R}^{5}, \mathrm{R}^{6}=\mathrm{H}, \mathrm{OH}$ $h ; R^{1}=R^{5}=R^{6}=H, R^{2}=R^{7}=O H, R^{3}=M e, R^{4}=O B n$ i; $R^{1}=H, R^{2}=R^{7}=O H, R^{3}=M e, R^{4}=O B n, R^{5}, R^{6}=H, O H$ j; $R^{1}=R^{7}=O H, R^{2}=R^{5}=R^{6}=H, R^{3}=M e, R^{4}=O B n$ $k ; R^{1}=R^{7}=O H, R^{2}=H, R^{3}=M e, R^{4}=O B n, R^{5}, R^{6}=H, O H$

(20)

(21)
product with sodium periodate in aqueous acetic acid solution resulted in disappearance of the starting material in 45 min and the appearance of a single product readily distinguished by TLC examination. The new compound was readily isolated by evaporation of the solution and crystallised from toluene as orange-red needles ( $98 \%$ yield), m.p. $250^{\circ} \mathrm{C}$. It is assigned the $O$-formyl structure (17b) which was confirmed by elemental analysis and mass [e.g. $\left.\left.m / z 490\left(M^{+}\right)\right], 269,91\left(100 \%, \mathrm{C}_{7} \mathrm{H}_{7}^{+}\right)\right]$ and ${ }^{1} \mathrm{H}$ NMR [e.g. $\delta 8.16$ (d, OCHO)] spectroscopy. The compound which is presumably an $\alpha, \beta$-mixture of anomers with sodium dithionite in aqueous sodium hydroxide at $0^{\circ} \mathrm{C}$ in a nitrogen atmosphere over 1-2 h produced a single main product which was isolated by aeration and acidification to give the anthracyclinone (19b) which crystallised from chloroform or toluene-ethyl acetate as orange-red needles, m.p. $209^{\circ} \mathrm{C}(90 \%$ yield). The structure was confirmed by elemental analysis, mass


Figure. CD spectra of some hydroxyglycitylquinizarins. $\mathrm{A}=(5 \mathrm{~S})-3-\mathrm{O}$ -Benzyl-3-C-methyl-1,2-O-isopropylidene-5-(9,10-dihydro-9,10-dioxo-2-anthryl)- $\alpha$-D-ribofuranose. $\mathrm{B}=(5 R)$-3- $O$-Benzyl-3-C-methyl-1,2-O-isopropylidene-5-(9,10-dihydro-9,10-dioxo- 2 -anthryl)- $\alpha$-D-ribofuranose. $\mathrm{C}=(5 R)$-3-O-Benzyl-3-C-methyl-1,2-O-isopropylidene-5-(9,10-dihydro-5-hydroxy-9,10-dioxo-2-anthryl)- $\alpha$-D-ribofuranose. $\mathrm{D}=(5 S)$ -3-C-Methyl-1,2-O-isopropylidene-5-(9,10-dihydro-9,10-dioxo-2-anthryl- $\alpha$-D-ribofuranose. See reference 1 for $C D$ spectra of some related hydroxyglycitylquinizarins.
[e.g. $\left.462\left(M^{+}\right), 444\left(M-H_{2} \mathrm{O}\right)\right]$ and ${ }^{1} \mathrm{H}$ NMR (e.g. absence of signal for $3^{\prime}-\mathrm{H}$ and assignment of all protons) spectroscopy.

Treatment of the anthracyclinone (19b) with boron trichloride in chloroform at $-78^{\circ} \mathrm{C}$ readily produced the corresponding debenzylated anthracyclinone (19a) the structure of which was confirmed by elemental analysis, mass [e.g. $m / z$ $\left.372\left(M^{+}\right)\right]$and ${ }^{1} H$ NMR (e.g. absence of signal for $3^{\prime}-\mathrm{H}$ and assignment of all protons) spectroscopy.

When the aldehyde (8b) was treated with leucoquinizarin in dimethylformamide (DMF) at $0^{\circ} \mathrm{C}$ in the presence of (DBU), two products were obtained namely the ( $5 R$ )-(9d) and ( $5 S$ )-(9c) hydroxyglycitylanthraquinones. The structure of the new ( $5 R$ ) isomer was confirmed by elemental analysis and mass [e.g. $m / z$ $\left.532\left(M^{+}\right)\right],{ }^{1} \mathrm{H}$ NMR, and CD spectroscopy.

In a similar series of reactions to that used for the ( $5 S$ ) isomer the ( $5 R$ ) isomer ( 9 d ) was converted into the pyranosyl derivative (15a) from which the anthracyclinone (19c) was obtained and the structure was confirmed by elemental analysis and mass [e.g. $\left.m / z 462\left(M^{+}\right)\right]$and ${ }^{1} \mathrm{H}$ NMR spectroscopy.

In preliminary experiments designed to produce anthracyclinones with additional functionality, especially hydroxy or alkoxy groups in ring A, we have examined the condensation of the $O$-benzyl aldehydo sugar ( 8 b ) with 5 -hydroxyleucoquinizarin (1b) in DMF in the presence of DBN. Three compounds were produced and readily separated by chromatography on silica gel and obtained in crystalline form. They are the $(5 R)(9 \mathrm{e}),(5 S)(9 f)$ hydroxyglycitylanthraquinones and the related 5 -deoxy derivative $(9 \mathrm{~g})$. The site of substitution in the anthraquinone is assigned on the basis of work on simple products (e.g. propyl 5-hydroxyquinizarin) produced in the normal Marschalk reaction in alkaline solution ${ }^{17}$ which indicates that substitution occurs in the $2^{\prime}$-position as shown, little of the 3 -substituted derivative being detectable. The ( $5 R$ ) and (5S) structures assigned were confirmed by comparison of CD spectra with those of known compounds (Figure). The $(5 R)$ isomer (9e) was readily deblocked by heating with $70 \%$ aqueous acetic acid to produce the crystalline pyranosyl anthraquinone ( $\mathbf{1 5 b}$ ). The structure of the compound was confirmed by elemental analysis and mass [e.g. m/z $508\left(M^{+}\right)$, $285\left(\mathrm{Q}^{\prime}: \mathrm{C}^{+} \mathrm{HOH} ; \mathrm{Q}^{\prime}=5\right.$-hydroxyquinizarin-2-yl) $]$ and ${ }^{1} \mathrm{H}$ NMR [e.g. coupling of $5-\mathrm{H}$ with only $4-\mathrm{H}(J 10 \mathrm{~Hz})$ and no evidence for $5-\mathrm{H}, 5-\mathrm{OH}$ coupling] spectra.

Reaction of the pyranosyl derivative (15b) with sodium metaperiodate in aqueous acetic acid was monitored by TLC and showed disappearance of starting material after 2.5 h at room temperature and appearance of a single new compound
which was readily isolated in crystalline form. It is assigned the $O$-formyl erythrose structure (18b) confirmed by elemental analysis, mass (e.g. $\left.\mathrm{m} / \mathrm{z} 506\left(M^{+}\right), 285\left(\mathrm{Q}^{\prime}: \mathrm{CHOH}\right)\right]$ and ${ }^{1} \mathrm{H}$ NMR (assignment of all proton) spectra and produced presumably from the first formed (16) by a $4-O$-formyl to $3-O-$ formyl group shift. When the latter compound (18b) was reduced with zinc in methanolic acetic acid it gave a yellow leuco derivative which with DBN in dry DMF gave two products in roughly equal amounts which were readily separated by chromatography on silica gel.

The ( $10 R$ ) hexahydroxy- and heptahydroxy-naphthacenedione structures (19h) and (19i) respectively may be assigned to the compounds. The structure of (19i) was confirmed by mass [e.g. $m / z 478\left(M^{+}\right), 460\left(M-H_{2} \mathrm{O}\right)$ ] and ${ }^{1} \mathrm{H}$ NMR (e.g. absence of signal for $3^{\prime}-\mathrm{H}$ and assignment of all protons) spectra. Similarly, the structure assigned to the hexahydroxy derivative (19h) was confirmed by mass and ${ }^{1} \mathrm{H}$ NMR spectroscopy. In an analogous manner the corresponding (10S) heptahydroxy (19k) and hexahydroxy (19j) naphthacenediones were prepared and characterised. The inhibitory effects of some of the anthracyclinones described herein and in earlier papers in this series on the proliferation of L1210/0, FM3A/0, Raji/0 Namalva and heptoma cells are outlined in the Table.

## Experimental

Evaporations were carried out under water-pump vacuum with a flask temperature below $40^{\circ} \mathrm{C}$ unless otherwise stated. UV spectra were measured with a Unicam SP800 spectrophotometer, IR spectra with a Perkin-Elmer 681 spectrophotometer, ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ) spectra with a JEOL JNM-MH-100 spectrometer (tetramethylsilane as internal standard) unless otherwise stated and mass spectra with an A.E.I. MS 903 spectrometer. We thank the SERC High Field NMR service for 400 MHz spectra and Dr. A. Drake, Birkbeck College, University of London for CD spectra. Silica gel ( $0.05-0.2 \mathrm{~mm}$, 70-270 mesh; Machery Nagel and Co.) was used for column chromatography. TLC was run on Silica Gel 60F 254 ( 0.2 mm thick) pre-coated aluminium plates (Merck) in the systems (A) toluene-ethyl acetate (4:1); (B) toluene-ethyl acetate (2:1).

## 3-C-Methyl-1,2:5,6-di-O-isopropylidene- $\alpha$-D-allofuranose

(5a).-The following relatively large-scale preparation was found satisfactory. ${ }^{13}$ A solution of 1,2:5,6-di- $O$-isopropylidene-$\alpha$-D-glucofuranose ( 250 g ), DMSO ( $2884 \mathrm{~cm}^{3}$ ), and acetic anhydride ( $1924 \mathrm{~cm}^{3}$ ) was set aside at room temperature for 24 h. TLC examination (system A) indicated absence of starting material ( $R_{\mathrm{F}} 0.30$ ) and appearance of a single product ( $R_{\mathrm{F}} 0.42$ ). The solution was evaporated at 0.05 mmHg to a syrup ( 273 g ). A solution of this ( 265 g ) in ether ( $1400 \mathrm{~cm}^{3}$ ) was added to methylmagnesium iodide [from methyl iodide ( $366 \mathrm{~cm}^{3}$ ) and magnesium ( 66.4 g )] in ether ( $1200 \mathrm{~cm}^{3}$ ) over 1.5 h with stirring. The solution was then boiled under reflux for 3 h , cooled, and washed with saturated aqueous ammonium chloride (1 200 $\mathrm{cm}^{\mathbf{3}}$ ). The aqueous phase was extracted twice with ether ( 1000 $\mathrm{cm}^{3}, 500 \mathrm{~cm}^{3}$ ) and the combined extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, and evaporated to leave a solid. The 3-C-methyl derivative ( 123 g) separated from light petroleum (b.p. $40-60^{\circ} \mathrm{C}$ ) as needles, m.p. $105-107^{\circ} \mathrm{C}$ (Found: 56.8; H, 7.8. Calc. for $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{6}$ : C, 56.9 ; $\mathrm{H}, 8.0 \%$ ).

3-C-Methyl-1,2-O-isopropylidene- $\alpha$-D-allofuranose (7a).-A solution of the foregoing di- $O$-isopropylidene derivative ( 8.7 g ) in $75 \%$ acetic acid $\left(70 \mathrm{~cm}^{3}\right)$ was set aside at room temperature for 24 h when TLC (system A) revealed complete disappearance of starting material and formation of a single new product ( $R_{F}$ 0.04 ). The solution was evaporated and coevaporated with

Table. Inhibitory effects of anthraquinone derivatives on the proliferation of L1210/0, FM3A/0, Raji/0, namalva and heptoma cells.*

| Compd. | $\mathrm{ID}_{50}(\mu \mathrm{~g} / \mathrm{ml})$ for cell growth (average values) |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | L1210/0 | FM3A/0 | Raji/0 | Namalva | Heptoma |
| (19a) | $4.82 \pm 1.01$ | $6.66 \pm 1.01$ | $4.00 \pm 1.07$ | $3.11 \pm 0.39$ | $4.18 \pm 0.79$ |
| (19b) | $99.5 \pm 38.5$ | $270 \pm 95$ | $36.3 \pm 6.9$ | $33.3 \pm 3.5$ | $92 \pm 49$ |
| (19c) | $78.6 \pm 10.7$ | 100 | $28.8 \pm 2.0$ | $29.3 \pm 0.8$ | $14.3 \pm 4.9$ |
| (19d) | $30.7 \pm 1.2$ | $29.6 \pm 7.5$ | $21.2 \pm 8.7$ | $11.0 \pm 5.8$ | $17.5 \pm 2.8$ |
| (19) | $10.4 \pm 4.65$ | $17.2 \pm 5.7$ | $11.2 \pm 6.05$ | $4.91 \pm 0.54$ | $3.67 \pm 0.19$ |
| (19) | $8.1 \pm 1.7$ | $15.7 \pm 7.07$ | $3.59 \pm 0.27$ | $2.80 \pm 0.09$ | $3.29 \pm 0.49$ |
| (19g) | 1000 | 1000 | 1000 | 1000 | $95.3 \pm 31.8$ |
| Adriamycin | $0.170 \pm 0.007$ |  |  |  |  |
| Daunomycin | $0.0173 \pm 0.0018$ |  |  |  |  |

* Antitumour test systems employed include the murine leukaemia L1210 cells, murine mammary carcinoma FM3A cells, human lymphoblastoid Namalva and Raji cells and rat Navikoff heptoma cells. The complete procedure for measuring inhibition of cell growth in vitro has been described elsewhere. ${ }^{18}$
toluene to provide a solid residue. The mono- $O$-isopropylidene derivative ( 7.2 g ) separated from chloroform-light petroleum (b.p. $\left.40-60^{\circ} \mathrm{C}\right)(1: 5)$ as needles, m.p. $133^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 51.1 ; \mathrm{H}$, 7.9. Calc. for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{O}_{6}$ : C, $51.3 ; \mathrm{H}, 7.7 \%$ ).
(5S)-3-C-Methyl-1,2-O-isopropylidene-5-(9',10'-dihydro-1', $4^{\prime}$-dihydroxy-9',10'-dioxo-2'-anthryl)- $\alpha$-D-ribofuranose (9a).To a solution of the foregoing diol (7a) (19.5g) in methanol (190 $\mathrm{cm}^{3}$ ) at $0{ }^{\circ} \mathrm{C}$ was added a solution of sodium metaperiodate $(18.5 \mathrm{~g})$ in water $\left(190 \mathrm{~cm}^{3}\right)$ over 30 min with stirring. After a further 60 min at $0^{\circ} \mathrm{C}$ ethanediol $\left(1 \mathrm{~cm}^{3}\right)$ was added. The filtered solution was evaporated under reduced pressure ( 0.05 mmHg ) to $c a .50 \mathrm{~cm}^{3}$, saturated with salt, and extracted with chloroform $\left(6 \times 50 \mathrm{~cm}^{3}\right)$. The organic layer was dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to give 3 - $C$-methyl-1,2- $O$-isopropylidene- $\alpha$-D-ribopentodialdehydo-1,4-furanose (8a) as a pale yellow syrup $(18.5 \mathrm{~g})$ which gave a single spot on TLC examination (system A) ( $R_{\mathrm{F}} 0.54$ ). It was used immediately in the next stage. To a solution of 2,3,4a,9a-tetrahydroanthracene-1,4,9,10-tetraone ( 26.6 g ) in THF ( $700 \mathrm{~cm}^{3}$ ) and methanol ( $700 \mathrm{~cm}^{3}$ ) at $-10^{\circ} \mathrm{C}$ was gradually added a solution of the foregoing dialdehydo derivative $(22.2 \mathrm{~g})$ in methanol $\left(100 \mathrm{~cm}^{3}\right)$ in an atmosphere of nitrogen over 20 min . A solution of $32 \%$ sodium hydroxide ( 25.8 $\mathrm{cm}^{3}$ ) was added and the mixture set aside for 45 min . Air was passed through the solution for 1.5 h and the resulting purple solution added to a mixture of $10 \mathrm{M} \mathrm{HCl}\left(100 \mathrm{~cm}^{3}\right)$, water ( 200 $\mathrm{cm}^{3}$ ), and crushed ice ( 200 g ) with rapid stirring. The resulting red solid precipitate ( 35 g ) was collected, washed with water, and dried in air. For chromatography a sample ( 20 g ) was dissolved in a mixture of toluene and ethyl acetate $(4: 1)$ and the filtered solution applied to a silica gel column $(10 \times 60 \mathrm{~cm})$. The major fraction ( $R_{\mathrm{F}} 0.58$ ), (system A) was readily eluted by toluene, ethyl acetate (4:1) and obtained crystalline by evaporation of the eluate. The title compound ( 19.2 g ) recrystallised from ethanol as orange-red needles, m.p. $203{ }^{\circ} \mathrm{C}$ (Found: C, 62.2; $\mathrm{H}, 5.1 \% ; M^{+}$, 442. $\mathrm{C}_{23} \mathrm{H}_{22} \mathrm{O}_{9}$ requires $\mathrm{C}, 62.45$; H, $5.0 \% ; M 442$ ); $m / z 442\left(M^{+}\right), 427\left(M-\mathrm{CH}_{3}\right), 269$, and 240 ; $v_{\text {max }}(\mathrm{KBr}) 3475(\mathrm{OH}), 1625$ and 1690 (quinizarin), and 1380 $\left(\mathrm{CMe}_{2}\right) \mathrm{cm}^{-1} ; \lambda_{\max }(\mathrm{EtOH})(\log \varepsilon) 482 \mathrm{~nm}(4.1) ; \delta\left(\mathrm{CDCl}_{3} ; 400\right.$ $\mathrm{MHz}) 1.3$ and $1.6(2 \times 3 \mathrm{H} \mathrm{CMe} 2), 2.62(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{OH}$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 2.90\left(1 \mathrm{H}, \mathrm{d}, J_{5,5-\mathrm{oH}} 3 \mathrm{~Hz}, 5-\mathrm{OH}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.2(2 \mathrm{H}, \mathrm{m}$, $2-\mathrm{H}, 4-\mathrm{H}), 5.3\left(1 \mathrm{H}, \mathrm{dd}, J_{5.5-\mathrm{oH}} 3 \mathrm{~Hz}, 5-\mathrm{H}\right), 5.84\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.5\right.$ $\mathrm{Hz}, 1-\mathrm{H}), 7.5\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right)$, $\mathrm{AA}^{\prime} \mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.80-7.90\left(6^{\prime}-7^{\prime}-\right.\right.$ H), $\delta_{\mathrm{B}} 8.32-8.40\left(5^{\prime}-\right.$ and $\left.\left.8^{\prime}-\mathrm{H}\right)\right], 12.83\left(1 \mathrm{H}, \mathrm{s}, 4^{\prime}-\mathrm{OH}\right.$ exch. $\mathrm{D}_{2} \mathrm{O}$ ), and 13.53 ( $1 \mathrm{H}, \mathrm{s}, 1^{\prime}-\mathrm{OH}$, exch. $\mathrm{D}_{2} \mathrm{O}$ ).

A solution of the compound $(0.5 \mathrm{~g})$ in pyridine $\left(5 \mathrm{~cm}^{3}\right)$ and acetic anhydride ( $50 \mathrm{~cm}^{3}$ ) and was left at room temperature for 24 h . The solution was evaporated to dryness and re-evaporated with toluene $\left(2 \times 10 \mathrm{~cm}^{3}\right)$ to give a yellow solid which was
washed with water $\left(100 \mathrm{~cm}^{3}\right)$ and shaken for 2 h with aqueous sodium hydrogen carbonate ( $100 \mathrm{~cm}^{3}$ ) then collected and washed with water $\left(2 \times 25 \mathrm{~cm}^{3}\right)$ and dried to give (5S)-3,5-di-O-acetyl-3-C-methyl-1,2-O-isopropylidene-5-( $1^{\prime}, 4^{\prime}$-diacetoxy$9^{\prime}, 10$-dihydro- $9^{\prime}, 10^{\prime}$-dioxo-2'-anthryl)- $\alpha$-D-ribofuranose (10a) which crystallised from toluene as yellow needles ( 0.5 g ), m.p. $159{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 60.6 ; \mathrm{H}, 4.8 \% ; M^{+}, 610 . \mathrm{C}_{31} \mathrm{H}_{30} \mathrm{O}_{13}$ requires C, 61.0 ; H, $4.95 \%$ M, 610 ); $m / z 610\left(M^{+}\right), 595\left(M-\mathrm{CH}_{3}\right), 558$ $\left(M-\mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}\right), 526,312(100 \%) ; v_{\max }(\mathrm{KBr}) 1735(\mathrm{CO})$ and $\left.1380 \mathrm{~cm}^{-1}\left(\mathrm{CMe}_{2}\right) ; \delta\left(\mathrm{CDCl}_{3}\right) ; 400 \mathrm{MHz}\right) 1.28-1.36(2 \times 3 \mathrm{H}$, $\mathrm{CMe}_{2}$ ), 1.5-1.65 ( $2 \times 3 \mathrm{H}, \mathrm{br}, \mathrm{CH}_{3}$ and 3-OCOMe), $2.1(3 \mathrm{H}, 5-$ OCOMe), 2.45 and $2.6\left(2 \times 3 \mathrm{H}, 1^{\prime}\right.$-OCOMe and $4^{\prime}$-OCOMe), $4.15(2 \mathrm{H}, 2-\mathrm{H}, 4-\mathrm{H}), 4.8(1 \mathrm{H}, \mathrm{br}, 5-\mathrm{H}), 5.85\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 3.5 \mathrm{~Hz}\right.$, $1-\mathrm{H}), 7.5\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), \mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.75-7.80\left(6^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}\right)\right.$, $\left.\delta_{\mathrm{B}} 8.15-8.20\left(5^{\prime}-\mathrm{H}, 8^{\prime}-\mathrm{H}\right)\right]$.
A solution of the tetra- $O$-acetate ( 0.2 g ) in $70 \%$ aqueous acetic acid ( $25 \mathrm{~cm}^{3}$ ) was boiled under reflux for 3 h . The solution was evaporated to dryness and re-evaporated with toluene to give a yellow solid; (5S)-3,5-di-O-acetyl-3-C-methyl-5-( $1^{\prime}, 4^{\prime}-$ diacetoxy- $9^{\prime}, 10^{\prime}$-dihydro- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$-anthryl)- $\alpha$-D-ribo-
furanose hemihydrate (20) $(0.18 \mathrm{~g})$ crystallised from tolueneethyl acetate as yellow needles, m.p. $>200^{\circ} \mathrm{C}$ (Found: C, 58.05; $\mathrm{H}, 4.6 \% ; M^{+}, 570 . \mathrm{C}_{28} \mathrm{H}_{26} \mathrm{O}_{13} \cdot \frac{.1}{2} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 58.05 ; \mathrm{H}$, $4.70 \% ; M, 570) ; m / z 570\left(M^{+}\right), 528$ ( $M-\mathrm{MeCO}$ ), 312; $\delta\left(\mathrm{CDCl}_{3} ; 400 \mathrm{MHz}\right) 1.22-1.25\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.33-1.55$ and 2.07-2.09 ( $2 \times 3 \mathrm{H}, 3-\mathrm{OCOMe}$ and $5-\mathrm{OCOMe}), 2.48-2.54$ $\left(2 \times 3 \mathrm{H}, 1^{\prime}\right.$-OCOMe, $4^{\prime}$-OCOMe), $3.84(1 \mathrm{H}, 1-\mathrm{OH}$ exch. $\mathrm{D}_{2} \mathrm{O}$ ), $4.12(2 \mathrm{H}, 2-\mathrm{H}, 4-\mathrm{H}), 5.44\left(1 \mathrm{H}, 2-\mathrm{OH}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 5.84$ $(1 \mathrm{H}, \mathrm{d}, J 3.5 \mathrm{~Hz}, 1-\mathrm{H}), 6.1(1 \mathrm{H}, \mathrm{br}, 5-\mathrm{H}), 7.52\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right)$, $\mathrm{AA}^{\prime} \mathrm{BB}^{\prime}$ signal $\delta_{\mathrm{A}} 7.72-7.78\left(6^{\prime}-\mathrm{H}\right.$ and $\left.7^{\prime}-\mathrm{H}\right), \delta_{\mathrm{B}} 8.14-8.18\left(5^{\prime}-\mathrm{H}\right.$, $\left.8^{\prime}-\mathrm{H}\right)$. A solution of the latter compound ( 0.2 g ) in acetic acid $\left(50 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$ was treated with a solution of sodium metaperiodate ( 0.071 g ) in water ( $50 \mathrm{~cm}^{3}$ ) and the mixture set aside for 1 h . Water ( $50 \mathrm{~cm}^{3}$ ) was added and the mixture extracted with chloroform ( $3 \times 25 \mathrm{~cm}^{3}$ ), the extract washed with water $\left(3 \times 25 \mathrm{~cm}^{3}\right)$ dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to a yellow solid. (4S,5S)-5-2,4-Di-O-acetyl-2-C-methyl-3-O-formyl-4-( $1^{\prime}, 4^{\prime}$-diacetoxy-9',10'-dihydro-9',10'-dioxo-2'-anthryl)-Daldehydoerythrose ( 21 ) ( 0.19 g ) crystallised from toluene as needles, m.p. $184{ }^{\circ} \mathrm{C} ; m / z 568\left(M^{+}\right), 553(M-\mathrm{Me}), 526(M-$ $\mathrm{MeCO}), 484(M-2 \mathrm{MeCO}) ; \delta_{\mathrm{H}} 1.2\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.55-$ $1.67\left(6 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCOCH}_{3}\right), 2.15\left(1 \times 3 \mathrm{H}, \mathrm{s}, 1^{\prime}-\mathrm{OCOCH}_{3}\right), 2.4$ $\left(1 \times 3 \mathrm{H}, \mathrm{s}, 4^{\prime}-\mathrm{OCOCH}_{3}\right), 4.1\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}_{4.3} 4 \mathrm{~Hz}, 4-\mathrm{H}\right.$ or $\left.3-\mathrm{H}\right), 4.5$ $\left(1 \mathrm{H}, \mathrm{d}, J_{3,4} 4 \mathrm{~Hz}, 3-\mathrm{H}\right.$ or 4 H$), 7.4\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), 7.8\left(2 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right.$ and $\left.7^{\prime}-\mathrm{H}\right), 8.15\left(2 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}\right.$ and $\left.8^{\prime}-\mathrm{H}\right), 8.2(1 \mathrm{H}, \mathrm{s}, \mathrm{OCHO})$, and 9.43 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{CCHO}$ ).

5-Oxo-3-C-methyl-1,2-O-isopropylidene-5-(9', $10^{\prime}$-dihydro$1^{\prime}, 4^{\prime}$-dihydroxy- $9^{\prime}, 10^{\prime}$-dioxo-2'-anthryl)- $\alpha$-D-ribofuranose (11).-

Pyridinium chlorochromate $(0.086 \mathrm{~g})$ was added to a solution of (5S)-3-C-methyl-1,2-O-isopropylidene-5-( $9^{\prime}, 10^{\prime}$-dihydro- $1^{\prime}, 4^{\prime}$ -dihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$-anthryl)- $\alpha$-D-ribofuranose ( 0.056 g ) in dry dichloromethane $\left(25 \mathrm{~cm}^{3}\right)$ and the mixture stirred at room temperature for 48 h when TLC examination (system B) revealed that the starting material had been replaced by a single new product. The solution was filtered, diluted with dichloromethane (to $c a .50 \mathrm{~cm}^{3}$ ), and washed with saturated aqueous sodium hydrogen carbonate ( $2 \times 200 \mathrm{~cm}^{3}$ ) and water until the aqueous wash was colourless. The extract was dried $\left(\mathbf{M g S O}_{4}\right)$, filtered through Celite, and evaporated to provide a red gum. This was dissolved in toluene-ethyl acetate $\left(4: 1 ; 30 \mathrm{~cm}^{3}\right)$ and the solution filtered through a pad of silica gel and evaporated to give a red solid. The 5-oxo-derivative ( 0.047 g ) crystallised from ethanol as red needles, m.p. $251{ }^{\circ} \mathrm{C}$; $v_{\max } 1715(\mathrm{CO})$ and $1380 \mathrm{~cm}^{-1}\left(\mathrm{CMe}_{2}\right) ; m / z 440\left(M^{+}\right), 425\left(M-\mathrm{CH}_{3}\right) 382$, and $267(100 \%) \delta\left(\mathrm{CDCl}_{3} ; 400 \mathrm{MHz}\right) 1.14-1.42\left(2 \times 3 \mathrm{H}, \mathrm{CMe}_{2}\right)$, $1.70\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.05\left(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.23(1 \mathrm{H}$, d, 2-H), $5.95(1 \mathrm{H}, \mathrm{d}, 1-\mathrm{H}), 5.75(1 \mathrm{H}, \mathrm{s}, 4-\mathrm{H}), 7.65\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right)$, $\mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.80-7.90\left(6^{\prime}-\mathrm{H}\right.\right.$ and $\left.7^{\prime}-\mathrm{H}\right), \delta_{\mathrm{B}} 8.32-8.45\left(5^{\prime}-\mathrm{H}\right.$ and $\left.\left.8^{\prime}-\mathrm{H}\right)\right], 12.83\left(1 \mathrm{H}, \mathrm{s}, 4^{\prime}-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$, and $13.85\left(1 \mathrm{H}, \mathrm{s}, 1^{\prime}-\right.$ OH exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(5S)-3-C-Methyl-5-(9',10'-dihydro- $1^{\prime}, 4^{\prime}$-dihydroxy- $9^{\prime}, 10-$
dioxo-2'-anthryl)- $\alpha$-( $\beta$ )-D-ribopyranose (14a).-A solution of the foregoing isopropylidene derivative (9a) ( 2.5 g ) in $70 \%$ aqueous acetic acid ( $250 \mathrm{~cm}^{3}$ ) was heated under reflux for 3 h when TLC examination (system A) showed complete disappearance of starting material ( $R_{\mathrm{F}} 0.58$ ) and the presence of a single product ( $R_{\mathrm{F}} 0.08$ ). The solution was evaporated to dryness and then re-evaporated with toluene. The title compound $(2.15 \mathrm{~g})$ separated from ethanol, or toluene as orange-red platelets, m.p. $240{ }^{\circ} \mathrm{C}$ (Found: C, 59.7; H, 4.6\%; $M^{+}, 402$. $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{O}_{9}$ requires C, $\left.59.7 ; \mathrm{H}, 4.5 \% ; M, 402\right) ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400\right.$ $\mathrm{MHz}], 1.4\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.90(1 \mathrm{H}, \mathrm{d}, 4-\mathrm{H} J 5 \mathrm{~Hz}), 2.55(1 \mathrm{H}, \mathrm{s}$, $2-\mathrm{H}), 4.50\left(1 \mathrm{H}, \mathrm{d}, 5-\mathrm{H}, J 5 \mathrm{H}_{3}\right), 5.39$ and $5.15(1 \mathrm{H}, \mathrm{d}$, signal split in the ratio $7: 1,1-\mathrm{H}), 6.45\left(1 \mathrm{H}, \mathrm{br}, 3-\mathrm{OH}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 6.70(1 \mathrm{H}$, $\mathrm{br}, 1-\mathrm{OH}$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.35\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right)$, and $\mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal [ $\delta_{\mathrm{A}} 7.80-7.90\left(6^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}\right), \delta_{\mathrm{B}} 8.05-8.15\left(5^{\prime}-\mathrm{H}\right.$ and $\left.8^{\prime}-\mathrm{H}\right)$ ], and 12.75-13.20 ( $2 \times 1 \mathrm{H}, \mathrm{br}, 1^{\prime}-\mathrm{OH}, 4^{\prime}-\mathrm{OH}$ exch. $\mathrm{D}_{2} \mathrm{O}$ ).

3-O-Benzyl-3-C-methyl-1,2:5,6-di-O-isopropylidene- $\alpha$-Dallofuranose ( $\mathbf{5 b}$ ).-A suspension of sodium hydride ( 2 g ) and 1,2:5,6-di- $O$-isopropylidene-3-C-methyl- $\alpha$-D-allofuranose (11 g) in dry DMSO ( $300 \mathrm{~cm}^{3}$ ) was stirred at $70^{\circ} \mathrm{C}$ for 1.5 h . Benzyl chloride ( 10.4 g ) was added gradually and the mixture heated at $70^{\circ} \mathrm{C}$ for 2 h . It was then poured into water $\left(100 \mathrm{~cm}^{3}\right)$ and crushed ice $(150 \mathrm{~g})$ and extracted with chloroform $\left(3 \times 50 \mathrm{~cm}^{3}\right)$. The combined extracts were washed with water ( $3 \times 50 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$, and evaporated to yield a crystalline solid. 3-O-Benzyl-3- C -methyl-1,2:5,6-di- $O$-isopropylidene- $\alpha$-D-allofuranose ${ }^{16}(13.2 \mathrm{~g})$ separated from hexane ( $100 \mathrm{~cm}^{3}$ ) as needles, m.p. $55-56{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 65.9 ; \mathrm{H}, 7.8 \% ; M^{+}, 364 . \mathrm{C}_{20} \mathrm{H}_{28} \mathrm{O}_{6}$ requires $\mathrm{C}, 65.9 ; \mathrm{H}, 7.7 \% ; M, 364) ; v_{\max }(\mathrm{KBr}) 1380 \mathrm{~cm}^{-1}$ $\left(\mathrm{CMe}_{2}\right)$.

3-O-Benzyl-3-C-methyl-1,2-O-isopropylidene- $\alpha$-D-allofuranose (7b).-A solution of the foregoing di-isopropylidene derivative ( 10 g ) in $70 \%$ aqueous acetic acid ( $200 \mathrm{~cm}^{3}$ ) was set aside at room temperature for 24 h when TLC examination revealed that the starting material had been replaced by a single new product. The solution was evaporated to dryness and reevaporated with toluene to give a solid residue. The monoisopropylidene derivative ( 8.6 g ) crystallised from ethanol as needles, m.p. $118-120^{\circ} \mathrm{C}$ (Found: C, 62.7; H, 7.5; $M^{+}, 317$. $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{O}_{6}$ requires C, $62.9 ; \mathrm{H}, 7.7 \% ; M, 317$ ); $v_{\text {max }} 1380 \mathrm{~cm}^{-1}$ ( $\mathrm{CMe}_{2}$ ).

3-O-Benzyl-1,2-O-isopropylidene-3-C-methyl- $\alpha$-D-ribopentodialdofuranose ( $\mathbf{8 b} \mathbf{)}$.-A solution of the foregoing allofuranose ( 10 g ) and sodium metaperiodate ( 7.0 g ) in aqueous acetic acid $\left(50 \% ; 60 \mathrm{~cm}^{3}\right)$ was set aside at room temperature for 24 h when the starting material had been replaced by a new product. The solution was diluted with water $\left(60 \mathrm{~cm}^{3}\right)$ and extracted with chloroform ( $4 \times 50 \mathrm{~cm}^{3}$ ) and the dried $\left(\mathrm{MgSO}_{4}\right)$ extract evaporated and co-evaporated with toluene to a solid residue. The furanose $(8.07 \mathrm{~g})$ crystallised from ethanol or toluene as needles, m.p. $40^{\circ} \mathrm{C}$ (Found: C, 65.5; H, 7.0\%; $M^{+}, 292$. $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{O}_{5}$ requires C, $65.7 ; \mathrm{H}, 6.9 \% ; \mathbf{M}, 292$ ); $\mathrm{v}_{\max }(\mathrm{KBr}) 1380$ ( $\mathrm{CMe}_{2}$ ) and $1735 \mathrm{~cm}^{-1}$ (CO).
(5S)-3-O-Benzyl-1,2-O-isopropylidene-3-C-methyl-5-(9',10'-dihydro-1', $4^{\prime}$-dihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$-anthryl)- $\alpha$-D-ribofuranose (9c).-A solution of the foregoing 3-O-benzyl-1,2-O-iso-propylidene-3- $C$-methyl- $\alpha$-D-ribopentodialdofuranose ( 6 g ) in methanol ( $40 \mathrm{~cm}^{3}$ ) was added with stirring to a solution of 2,3,4a,9a-tetrahydroanthracene-1,4,9,10-tetraone ( 4.93 g ) in THF ( $150 \mathrm{~cm}^{3}$ ) and methanol $\left(100 \mathrm{~cm}^{3}\right)$ at $-10^{\circ} \mathrm{C}$ in a nitrogen atmosphere. After 15 min , a solution of sodium hydroxide ( $10 \%$; $10 \mathrm{~cm}^{3}$ ) was added. After 1 h a major product ( $R_{\mathrm{F}} 0.68$, system A) was produced in addition to unchanged quinizarin ( $R_{\mathrm{F}} 0.98$ ) and a minor product ( $R_{\mathrm{F}} 0.88$ ). The solution was aerated for 1.5 $h$ then added gradually to a mixture of $10 \mathrm{~m}-\mathrm{HCl},\left(20 \mathrm{~cm}^{3}\right)$, water ( $20 \mathrm{~cm}^{3}$ ), and crushed ice ( 40 g ) with rapid stirring. The resultant orange-red solid precipitate ( 8.5 g ) was collected, washed with water ( $2 \times 20 \mathrm{~cm}^{3}$ ), and air-dried. A filtered solution of the solid in toluene-ethyl acetate ( $4: 1$ ) was applied to a silica gel column $(7.5 \times 32 \mathrm{~cm})$ and eluted by the same solvent system. The major product ( $R_{\mathrm{F}} \mathbf{0} \mathbf{0 . 6 8}$, system A ) readily separated from quinizarin and the minor product and was obtained as a red solid after evaporation of the appropriate column fractions. The title compound ( 4.1 g ) crystallised from ethanol as fine orange-red needles, m.p. $189^{\circ} \mathrm{C}$ which retained a little water (Found: C, 67.1; H, $5.3 \% ; M^{+}, 532 . \mathrm{C}_{30} \mathrm{H}_{28} \mathrm{O}_{9}, 1 / 3$ $\mathrm{H}_{2} \mathrm{O}$ requires C, 66.9; $\mathrm{H}, 5.2 \% ; M, 532$ ); $m / z 514\left(M-\mathrm{H}_{2} \mathrm{O}\right)$ 269, 240, and $91\left(\mathrm{C}_{7} \mathrm{H}_{7}^{+}, 100 \%\right) ; v_{\max }(\mathrm{KBr}) 1380\left(\mathrm{CMe}_{2}\right) \mathrm{cm}^{-1}$; $\delta\left(\mathrm{CDCl}_{3} ; 100 \mathrm{MHz}\right) 1.32,1.40,1.56\left(2 \times 3 \mathrm{H}, \mathrm{CMe}_{2}\right.$ and $1 \times 3$ $\mathrm{H}, \mathrm{C}-\mathrm{Me})$, $2.92\left(1 \mathrm{H}, \mathrm{br}, 5-\mathrm{OH}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.36\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}_{2,1} 3\right.$ $\mathrm{Hz}, 2-\mathrm{H}), 4.44\left(1 \mathrm{H}, \mathrm{d}, J_{4,5} 7.5 \mathrm{H}_{3}, 4-\mathrm{H}\right), 4.83(2 \times 1 \mathrm{H}, \mathrm{d}, J 10$ $\left.\mathrm{Hz}, \mathrm{OCH}_{2} \mathrm{Ph}\right), 5.48\left(1 \mathrm{H}, \mathrm{d}, J_{5,4} 7.5 \mathrm{~Hz}, 5-\mathrm{H}\right), 5.80\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 3\right.$ $\mathrm{Hz}, 1-\mathrm{H}), 7.0\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.44\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), \mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal ( $\delta_{\mathrm{A}} 7.68-7.88\left(6^{\prime}\right.$ - and $\left.7^{\prime}-\mathrm{H}\right), \delta_{\mathrm{B}} 8.16-8.40\left(5^{\prime}-\right.$ and $\left.\left.8^{\prime}-\mathrm{H}\right)\right], 12.72$ $\left(1 \mathrm{H}, \mathrm{s}, 4^{\prime}-\mathrm{OH}\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$ ), and $13.44\left(1 \mathrm{H}, \mathrm{s}, 1^{\prime}-\mathrm{OH}\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$ ).

A portion of the compcund ( 0.2 g ) in pyridine ( $2 \mathrm{~cm}^{3}$ ) and acetic anhydride ( $20 \mathrm{~cm}^{3}$ ) was set aside for 24 h at room temperature when TLC examination (system A) revealed that the starting material ( $R_{F} 0.68$ ) had been replaced by a single product ( $R_{\mathrm{F}} 0.86$ ). The solution was added to a mixture of ice and water ( $30 \mathrm{~cm}^{3}$ ) and extracted with chloroform ( $3 \times 20$ $\mathrm{cm}^{3}$ ). The combined extracts were washed with water ( $3 \times 25$ $\mathrm{cm}^{3}$ ), dried ( $\mathrm{MgSO}_{4}$ ), and evaporated to a yellow solid.
(5S)-5-O-Acetyl-3-O-benzyl-3-C-methyl-1,2-O-isopropylid-ene-5-( $1^{\prime}, 4^{\prime}$-diacetoxy- $9^{\prime}, 10^{\prime}$-dihydro- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$-anthryl $)-\alpha$ -D-ribofuranose hemihydrate ( 10 b ) $(0.2 \mathrm{~g})$ crystallised from ethanol as yellow needles, m.p. $187^{\circ} \mathrm{C}$ (Found: C, $64.4 ; \mathrm{H}, 5.1 \%$;
 658); $m / z 658\left(M^{+}\right), 643(M-M e), 325,323,311$ (and 91 $\left(100 \%, \mathrm{C}_{7} \mathrm{H}_{7}^{+}\right) ;\left(\delta \mathrm{CDCl}_{3} ; 400 \mathrm{MHz}\right) 1.27-1.34(2 \times 3 \mathrm{H}$, $\mathrm{CMe}_{2}$ ), $1.56-1.64(1 \times 3 \mathrm{H}, \mathrm{Me}), 2.03-2.12(3 \mathrm{H}, 5-\mathrm{OCOMe})$, $2.42-2.50\left(2 \times 3 \mathrm{H}, 1^{\prime}-\mathrm{OCOMe}\right.$ and $\left.4^{\prime}-\mathrm{OCOMe}\right), 4.34(2 \mathrm{H}$, $2-\mathrm{H}, 4-\mathrm{H}), 4.44(1 \mathrm{H}, \mathrm{br}, 5-\mathrm{H})$, $5.82\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 3 \mathrm{~Hz}\right.$, $1-\mathrm{H}), 7.0-7.20\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.51\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right)$, $\mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}-$ signal $\left[\delta_{\mathrm{A}} 7.72-7.79\left(6^{\prime}-\mathrm{H}\right.\right.$ and $\left.7^{\prime}-\mathrm{H}\right), \delta_{\mathrm{B}} 8.10-8.18\left(5^{\prime}-\mathrm{H}\right.$ and $\left.8^{\prime}-\mathrm{H}\right)$ ].
(5R)-3-O-Benzyl-1,2-O-isopropylidene-3-C-methyl-5-(9',10'-dihydro-1', $4^{\prime}$-dihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$-anthryl)- $\alpha$-D-ribofuranose (9d).-To a solution of 2,3,4a,9a-tetrahydroanthracene-$1,4,9,10$-tetraone ( 5 g ) in dry DMF ( $40 \mathrm{~cm}^{3}$ ) at $0^{\circ} \mathrm{C}$ was added a solution of 3-O-benzyl-1,2- $O$-isopropylidene-3-C-methyl- $\alpha-$ D-ribopentodialdofuranose ( 7.1 g ) in DMF ( $10 \mathrm{~cm}^{3}$ ) and the mixture kept under nitrogen for 15 min ; it was then treated with DBU ( $7.5 \mathrm{~cm}^{3}$ ) and left under nitrogen for 15 min . TLC examination of the solution showed two major products ( $R_{F}$ 0.68 ) and ( $R_{\mathrm{F}} 0.88$ ) plus a little leucoquinizarin ( $R_{\mathrm{F}} 0.98$ ). A steady stream of air was passed through the solution for 25 min and the resultant purple solution added gradually to a mixture of crushed ice and $2 \mathrm{~m}-\mathrm{HCl}\left(120 \mathrm{~cm}^{3}\right)$ with rapid stirring. A red solid precipitate was extracted into toluene-ethyl acetate (1:1; $3 \times 100 \mathrm{~cm}^{3}$ ) and the extract evaporated and then coevaporated with toluene to provide a red solid. A filtered solution of the solid in toluene-ethyl acetate (4:1) was applied to a silica gel column $(7.5 \times 40 \mathrm{~cm})$ and eluted by the same solvent mixture.

Two major products were obtained, compound A (4g) ( $R_{\mathrm{F}}$ $0.88)$ and compound $B(2 \mathrm{~g})\left(R_{F} 0.68\right)$ in system $A$. The latter compound $B$ was identical with the ( $5 S$ ) derivative prepared above. Compound A, the (5R) derivative (9d) crystallised from ethanol as orange-red needles, m.p. $127^{\circ} \mathrm{C}$ which retained a little water (Found: C, 67.2; H, 5.4\%; $M^{+}, 532 . \mathrm{C}_{30} \mathrm{H}_{28} \mathrm{O}_{9}-\frac{1}{4} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 67.1 ; \mathrm{H}, 5.2 \%, M, 532$ ); $\lambda_{\text {max }}(\mathrm{EtOH}) 520,487,328$, and $248 \mathrm{~nm} ; v_{\max }(\mathrm{KBr}) 3500(\mathrm{OH}), 1625,1690$ (quinone), and $1380\left(\mathrm{CMe}_{2}\right) \mathrm{cm}^{-1} ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}, 400 \mathrm{MHz}\right] 1.25(1 \times 3 \mathrm{H}, \mathrm{s}$, $\mathrm{CH}_{3}$ ), 1.35 and $1.45\left(2 \times 3 \mathrm{H}, \mathrm{CMe}_{2}\right), 4.36\left(2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.44(1$ $\left.\mathrm{H}, \mathrm{d}, J_{2.1} 4 \mathrm{~Hz}, 2-\mathrm{H}\right), 4.5(1 \mathrm{H}, \mathrm{d}, 4-\mathrm{H}), 5.27\left(1 \mathrm{H}, \mathrm{d}, J_{5, \text { - } \mathrm{OH}} 5.5\right.$ $\mathrm{Hz}, 5-\mathrm{H}), 5.70\left(1 \mathrm{H}, \mathrm{br}, J_{5-\mathrm{or}, 5} 5.5 \mathrm{~Hz}, 5-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 5.75(1$ $\left.\mathrm{H}, \mathrm{d}, J_{1,2} 4 \mathrm{~Hz}, 1-\mathrm{H}\right), 7.0\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.56\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), \mathrm{AA}^{\prime}-$ $\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.97-8.00\left(6^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}\right), \delta_{\mathrm{B}} 8.24-8.29\left(5^{\prime}-\mathrm{H}\right.\right.$ and $8^{\prime}-$ H) ], $12.80\left(1 \mathrm{H}, \mathrm{s}, 4^{\prime}-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$, and $13.4\left(1 \mathrm{H}, \mathrm{s}, 1^{\prime}-\mathrm{OH}\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(5R)-3-O-Benzyl-3-C-methyl-5-(9',10'-dihydro-1',4'-di-hydroxy- $9^{\prime}, 10^{\prime}$-dioxo-2'-anthryl)- $\alpha$-( $\beta$ )-D-ribopyranose (15a).A solution of the foregoing ( $5 R$ )-isopropylidene derivative ( 1.0 g ) in $70 \%$ aqueous acetic acid ( $150 \mathrm{~cm}^{3}$ ) was boiled under reflux for 2.5 h when TLC examination (system A) revealed complete disappearance of starting material ( $R_{\mathrm{F}} 0.68$ ) and its replacement by a single product ( $R_{\mathrm{F}} 0.40$ ). The solution was evaporated to dryness and then co-evaporated with toluene to give a red solid $(0.92 \mathrm{~g})$. The O-benzyl derivative monohydrate ( $\mathbf{1 4 \mathrm { b } ) \text { crystallised }}$ from ethanol as orange-red plates, m.p. $240{ }^{\circ} \mathrm{C}$ (Found: C, 63.8; $\mathrm{H}, 5.0 \% \cdot \mathrm{C}_{27} \mathrm{H}_{24} \mathrm{O}_{9} \cdot \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 63.5 ; \mathrm{H}, 5.15 \%$; $\delta$ $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right], 1.5\left(1 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.15\left(1 \mathrm{H}, \mathrm{d}, J_{2,1}\right.$ $8 \mathrm{~Hz}, 2-\mathrm{H}), 3.37\left(1 \mathrm{H}, \mathrm{br}, 1-\mathrm{OH}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 3.60\left(1 \mathrm{H}, \mathrm{d}, J_{4,5}\right.$ $10 \mathrm{~Hz}, 4-\mathrm{H}), 4.83\left(2 \times 1 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.95\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 8 \mathrm{~Hz}\right.$, $1-\mathrm{H}), 5.08\left(1 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{H}_{3}, 2-\mathrm{OH}\right.$ exch. $\mathrm{D}_{2} \mathrm{O}$ ), $5.29(1 \mathrm{H}, \mathrm{d}$, $\left.J_{5.4} 10 \mathrm{~Hz}, 5-\mathrm{H}\right), 6.76\left(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, 4-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.25-$ $7.50\left(6 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}\right.$ and $\left.\mathrm{C}_{6} \mathrm{H}_{5}\right), \mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.93-7.98\right.$ $\left.\left(6^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}\right), \delta_{\mathrm{B}} 8.22-8.27\left(5^{\prime}-\mathrm{H}, 8^{\prime}-\mathrm{H}\right)\right], 12.8,13.3(2 \times 1 \mathrm{H}$, $1^{\prime}-\mathrm{OH}, 4^{\prime}-\mathrm{OH}$ each exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(5S)-3-O-Benzyl-3-C-Methyl-5-(9',10'-dihydro-1', $4^{\prime}$-di-hydroxy-9',10'-dioxo-2'-anthryl)- $\alpha, \beta$-D-ribopyranose (14b).-A solution of the foregoing ( $5 S$ )-isopropylidene derivative ( 1.005 g) in $70 \%$ aqueous acetic acid ( $150 \mathrm{~cm}^{3}$ ) was boiled under reflux for 2.5 h when TLC (system A) revealed the complete disappearance of starting material ( $R_{\mathrm{F}} 0.88$ ) and its replacement by a single product ( $R_{\mathrm{F}} 0.44$ ). The solution was evaporated to dryness and re-evaporated with toluene to give the title compound (14b) ( 0.93 g ), m.p. $145{ }^{\circ} \mathrm{C}$ (Found: C, 66.0; H, 5.0\%; $M^{+}, 492 \mathrm{C}_{27} \mathrm{H}_{24} \mathrm{O}_{9}$ requires $\mathrm{C}, 65.85 ; \mathrm{H}, 5.0 \% ; \mathrm{M}, 492$ ); $\lambda_{\max }(\mathrm{EtOH}) 532,519,486$, and $329 ; \nu_{\max }(\mathrm{KBr}) 3500(\mathrm{OH})$ and 1625,1690 (quinone) $\mathrm{cm}^{-1} ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.52(1$
$\mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}$ ), $3.1\left(1 \mathrm{H}, \mathrm{d}, J_{4-\mathrm{OH}, 4-\mathrm{H}} 6.5 \mathrm{~Hz}, 4-\mathrm{OH}\right.$ exch. $\mathrm{D}_{2} \mathrm{O}$ ), 3.55 ( $1 \mathrm{H}, \mathrm{d}, J_{2 \text {-OH2,-H }} 7.5 \mathrm{~Hz}, 2-\mathrm{OH}$, exch. $\mathrm{D}_{2} \mathrm{O}$ ), $4.1\left(1 \mathrm{H}, \mathrm{d}, J_{2,2 \text {-OH }}\right.$ $\left.7.5 \mathrm{H}_{3}, 2-\mathrm{H}\right)$, $4.15\left(1 \mathrm{H}, \mathrm{d}, J_{1,10 \mathrm{H}} 12 \mathrm{H}_{3}, 1-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.4$ $\left(1 \mathrm{H}, \mathrm{d}, J_{4,4-\mathrm{OH}} 6.5 \mathrm{H}_{3} 4-\mathrm{H}\right), 4.8\left(2 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.3(1 \mathrm{H}, \mathrm{d}$, $\left.J_{1,10 \mathrm{H}} 12 \mathrm{H}_{3}, 1-\mathrm{H}\right), 5.45(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{H}), 7.35(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 7.5(1$ $\mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}$ ), an $\mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.95-8.05\right.$ ( $6^{\prime}$ and $7^{\prime}-\mathrm{H}$ ); $\delta_{\mathrm{B}}$ 8.33-8.38 ( $5^{\prime}$-and $\left.8^{\prime}-\mathrm{H}\right)$ and 12.8 and $13.4\left(2 \times 1 \mathrm{H}, 1^{\prime}\right.$ and $4^{\prime}-\mathrm{H}$ both exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(4R)-2-C-Methyl-2-O-benzyl-3-O-formyl-4-(9',10'-dihydro$1^{\prime}, 4^{\prime}$-dihydroxy- $9^{\prime}, 10^{\prime}$-dioxo-2'-anthryl)-D-erythro-tetrafuranose (18a).-To a solution of the foregoing ( $5 R$ )-3-O-benzyl-3-$C$-methyl- $\alpha, \beta$-d- 5 -( $9^{\prime}, 10^{\prime}$-dihydro- $1^{\prime}, 4^{\prime}$-dihydroxy- $9^{\prime}, 10^{\prime}$-dioxo-$2^{\prime}$-anthryl)-ribopyranose ( 0.93 g ) in $70 \%$ aqueous acetic acid ( $150 \mathrm{~cm}^{3}$ ) was added a solution of sodium metaperiodate ( 0.432 g ) in water ( $50 \mathrm{~cm}^{3}$ ) and acetone ( $20 \mathrm{~cm}^{3}$ ) and the mixture stirred at room temperature for 5.5 h when TLC examination (system A) revealed the absence of starting material and the presence of major ( $R_{\mathrm{F}} 0.90$ ) and minor ( $R_{\mathrm{F}} 0.62$ ) products. Water ( $100 \mathrm{~cm}^{3}$ ) was added and the mixture extracted with chloroform ( $3 \times 50 \mathrm{~cm}^{3}$ ). The extract was washed with water ( $3 \times 25 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to a red solid. The ( $4 R$ )-erythro-tetrafuranose ( $\mathbf{1 8 a}$ ) ( 0.9 g ) crystallised from toluene as orange-red needles, m.p. $>200^{\circ} \mathrm{C}$ (Found: C, 65.8 ; $\mathrm{H}, 4.9 \% ; M^{+}, 490 . \mathrm{C}_{27} \mathrm{H}_{22} \mathrm{O}_{9}$ requires $\mathrm{C}, 66.1 ; \mathrm{H}, 4.5 \% ; M$, 490); $m / z 490\left(M^{+}\right), 269(\mathrm{Q}=\mathrm{CHOH}), 267\left(\mathrm{QCO}^{+}\right), 254,240$, $91\left(100 \%, \mathrm{C}_{7} \mathrm{H}_{7}^{+}\right), \lambda_{\max }(\mathrm{EtOH}) 520,481$, and 329 nm ; $\delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.50-1.56(3 \mathrm{H}$, s split in ratio $3: 1$, $\mathrm{CH}_{3}$ ), $3.60\left(1 \mathrm{H}, \mathrm{d}, 1-\mathrm{OH}\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$ ), $4.50-4.60(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{2} \mathrm{Ph}$ ), $5.3\left(1 \mathrm{H}, \mathrm{d}, 1-\mathrm{H}\right.$, collapses to a singlet with $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.20-$ $7.36(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 7.48\left(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-3^{\prime}\right), \mathrm{AA}^{\prime}-\mathrm{BB}^{\prime} \operatorname{signal}\left[\delta_{\mathrm{A}} 7.90\left(6^{\prime}-\right.\right.$ $\left.\left.\mathrm{H}, 7^{\prime}-\mathrm{H}\right), \delta_{\mathrm{B}} 8.30\left(5^{\prime}-\mathrm{H}, 8^{\prime}-\mathrm{H}\right)\right] 8.15(\mathrm{H}, \mathrm{d}, \mathrm{OCHO}), 12.6-13.2$ $\left(2 \times 1 \mathrm{H}, \mathrm{br}, 1^{\prime}-\mathrm{OH}, 4^{\prime}-\mathrm{OH}\right.$ both exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$.
(7R,S,8R,9S,10R)-8-Benzyloxy-8-methyl-7,8,9,10-tetrahydro-6,7,9,10,11-pentahydroxynaphthacene-5,12-dione (19b).-A solution of the foregoing ( 4 R )- O -benzyl erythrose derivative ( $\mathbf{1 6 \mathrm { b } \text { ) } ) ~ ( ~}$ $(0.7 \mathrm{~g})$ in $7 \%$ aqueous sodium hydroxide $\left(50 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$ was stirred for 1.5 h in a nitrogen atmosphere and then treated with a solution of sodium dithionite ( 0.55 g ) in $7 \%$ aqueous sodium hydroxide ( $10 \mathrm{~cm}^{3}$ ). After 1.5 h the yellow-brown solution was aerated for 2 h to produce a purple solution. This was added gradually to $5 \mathrm{M}-\mathrm{HCl}\left(40 \mathrm{~cm}^{3}\right)$ and ice ( 40 g ). The resultant red solid precipitate was collected, washed with water, and dried. The anthracyclinone hemihydrate ( 0.6 g ) crystallised from chloroform or toluene-ethyl acetate as orange-red needles, m.p. $209{ }^{\circ} \mathrm{C}$ (Found: C, 66.4; H, $4.9 \% ; M^{+}$, 462. $\mathrm{C}_{25} \mathrm{H}_{22} \mathrm{O}_{8} \cdot \frac{1}{2} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 66.25 ; \mathrm{H}, 4.9 \% ; M, 462) ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right]$ $1.5\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 4.0(1 \mathrm{H}, \mathrm{t}, 7-\mathrm{H}), 4.53,4.72\left(2 \times 1 \mathrm{H}, \mathrm{d}, J_{9,10}\right.$ $12.5 \mathrm{~Hz}, 10-\mathrm{H}, 9-\mathrm{H}), 4.79\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.23(1 \mathrm{H}, \mathrm{br}, 7-\mathrm{OH}$ exch. $\mathrm{D}_{2} \mathrm{O}$ ), $5.42\left(1 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 9-\mathrm{OH}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 6.15(1$ $\mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 10-\mathrm{OH}$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.10\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), \mathrm{AA}^{\prime}-$ $\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}}(7.82-7.90, \mathrm{H}-2, \mathrm{H}-3), \delta_{\mathrm{B}}(8.32-8.36,1-\mathrm{H}, 4-\mathrm{H})\right]$ 13.3 and $13.8\left(2 \times 1 \mathrm{H}, \mathrm{s}, 6\right.$-and 11-OH both exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$.
(4S)-2-C-Methyl-2-O-benzyl-3-O-formyl-4-(9',10'-dihydro$1^{\prime}, 4^{\prime}$-dihydroxy-9,10-dioxo-2-anthryl)- $\alpha, \beta$-D-erythro-tetrafuranose (17b).-(a) To a solution of the foregoing (5S)-pyranose derivative ( 0.92 g ) in acetic acid ( $150 \mathrm{~cm}^{3}$ ) was added a solution of sodium metaperiodate ( 0.432 g ) in water ( $100 \mathrm{~cm}^{3}$ ). The starting material was completely replaced by a single new product ( $R_{\mathrm{F}} 0.82$, system A). After 45 min water ( $200 \mathrm{~cm}^{3}$ ) was added and the mixture extracted with chloroform ( $4 \times 50$ $\mathrm{cm}^{3}$ ). The combined extracts were washed with water, dried ( $\mathrm{MgSO}_{4}$ ) and evaporated to provide a red solid. The erythrose derivative $(0.9 \mathrm{~g})$ crystallised from toluene as orange-red needles, m.p. $250^{\circ} \mathrm{C}$ (Found: C, 65.4; H, 4.6\%; $M^{+}, 490$. $\mathrm{C}_{27} \mathrm{H}_{22} \mathrm{O}_{9} \cdot 1 / 3 \quad \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 65.3 ; 4.95 \% ; M, 490$ ); $\delta$ -
[ $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right]$ 1.45-1.5 ( 3 H , singlets split in ratio $3: 2$, $\left.\mathrm{CH}_{3}\right), 3.97\left(1 \mathrm{H}, \mathrm{d}, J_{1-\mathrm{OH}, 1} 8 \mathrm{~Hz}, 1-\mathrm{OH}\right.$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.55-4.75$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}$ ), $5.39\left(1 \mathrm{H}, \mathrm{d}, J_{1, \mathrm{OH}-1} 8 \mathrm{~Hz}\right.$, collapses to a singlet with $\left.\mathrm{D}_{2} \mathrm{O}, 1 \mathrm{H}\right), 7.20\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), 7.3-7.45\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right)$, $\mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.05\left(6^{\prime}-\mathrm{H}, 7-\mathrm{H}\right), \delta_{\mathrm{B}} 8.35\left(5^{\prime} \mathrm{H}, 8^{\prime} \mathrm{H}\right)\right]$, $8.16(1 \mathrm{H}, \mathrm{d}, \mathrm{OCHO}), 12.85$ and $13.60\left(2 \times 1 \mathrm{H}, 1-\mathrm{OH}, 4^{\prime}-\mathrm{OH}\right.$ each exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(7R/S,8R,9S,10S)-8-O-Benzyl-8-methyl-7,8,9,10-tetrahydro-6,7,9,10,11-pentahydroxynaphthacene-5,12-dione (19b).-To a solution of the ( $4 S$ ) $O$-benzylerythrose ( 16 c ) $(0.9 \mathrm{~g})$ in $7 \%$ aqueous sodium hydroxide ( $60 \mathrm{~cm}^{3}$ ) at $0^{\circ} \mathrm{C}$ in a nitrogen atmosphere was added sodium dithionite ( 0.75 g ) in $7 \%$ aqueous sodium hydroxide ( $15 \mathrm{~cm}^{3}$ ). After 2 h the starting material had been replaced by one major product ( $R_{\mathrm{F}} 0.60$ ) (system A). Air was passed through the solution for 2.5 h and the resultant purple solution was added to $2 \mathrm{M}-\mathrm{HCl}\left(40 \mathrm{~cm}^{3}\right)$ and crushed ice ( 20 g ) with stirring. The red solid precipitate was collected and washed with water. The (10S)-anthracyclinone (19b) ( 0.6 g ) crystallised from toluene-ethyl acetate ( $1: 1$ ) or chloroform as orange-red needles, m.p. $121^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 67.4 ; \mathrm{H}, 4.9 \% ; M^{+}$, 462. $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{O}_{8}$ requires $\left.\mathrm{C}, 67.5 ; \mathrm{H}, 4.8 \% ; M, 462\right) ; \lambda_{\max }(\mathrm{EtOH})$ 520,484 , and $336 \mathrm{~nm} \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.5(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CH}_{3}\right), 5.23,5.42,6.15(3 \times 1 \mathrm{H}, 7-, 9$-, and $10-\mathrm{OH}$ all exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.0\left(1 \mathrm{H}, \mathrm{t}\right.$, collapses to doublet with $\left.\mathrm{D}_{2} \mathrm{O}, J 5 \mathrm{H}_{3}, 9-\mathrm{H}\right)$, 4.53 and $4.72\left(2 \times 1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{H}_{3}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.79\left(1 \mathrm{H}, \mathrm{d}, J_{10.9}\right.$ $\left.5 \mathrm{H}_{3}, 10-\mathrm{H}\right), 4.8(1 \mathrm{H}, \mathrm{s}, 7-\mathrm{H}), 7.10(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph})$, an $\mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.82-7.90(2-\right.$ and $3-\mathrm{H}), \delta_{\mathrm{B}} 8.29-8.35(1-$ and $4-\mathrm{H})$ ], 13.3 and $13.6\left(1 \times 2 \mathrm{H}, \mathrm{s}, 6\right.$ - and $11-\mathrm{OH}$ both exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$.
(7R,S,8R,9S,10S)-8-Methyl-7,8,9,10-tetrahydro-6,7,8,9,10,11-hexahydroxynaphthacene-5,12-dione (19c).-(a) The foregoing $10(S)$-anthracyclinone ( 50 mg ) in chloroform ( $25 \mathrm{~cm}^{3}$ ) was treated with boron trichloride at $-78^{\circ} \mathrm{C}$ and then set aside for 10 min when the temperature of the solution was $0^{\circ} \mathrm{C}$. Water was added and the mixture evaporated to dryness. The residue was treated with water to give a red precipitate which was collected. The anthracyclinone ( 35 mg ) formed an orange-red solid homogeneous on TLC ( $R_{\mathrm{F}} 0.06$, system A) (Found: $M^{+}$, $372 \mathrm{C}_{19} \mathrm{H}_{16} \mathrm{O}_{8}$ require $M, 372 ; \lambda_{\max }(\mathrm{EtOH}) 519,485$, and 330 $\mathrm{nm} ; \delta\left(\mathrm{CDCl}_{3}, 400 \mathrm{mHz}\right) 1.5\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.1(1 \mathrm{H}, \mathrm{br}, 8-\mathrm{OH}$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 2.70-2.88(2 \times 1 \mathrm{H}, \mathrm{br}, 9-\mathrm{OH}, 7-\mathrm{OH}$ each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.40\left(1 \mathrm{H}, \mathrm{br}, 10-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.20\left(1 \mathrm{H}, \mathrm{d}, J_{9,10} 7 \mathrm{~Hz}\right.$, $9-\mathrm{H}), 4.91(1 \mathrm{H}, 9-\mathrm{H}), 5.06\left(1 \mathrm{H}, \mathrm{d}, J_{10.9} 7 \mathrm{~Hz}, 10-\mathrm{H}\right), \mathrm{AA}^{\prime}-\mathrm{BB}^{\prime}$ signal $\left[\delta_{\mathrm{A}} 7.85-7.89(2-\mathrm{H}, 3-\mathrm{H}), \delta_{\mathrm{B}} 8.34-8.38(1-\mathrm{H}, 4-\mathrm{H})\right], 13.45$ and $13.80\left(2 \times 1 \mathrm{H}, \mathrm{br}, 6-\mathrm{OH}, 11-\mathrm{OH}\right.$ each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$.
(b) To a solution of the ribopyranose (14a) ( 0.5 g ) in $70 \%$ aqueous acetic acid $\left(75 \mathrm{~cm}^{3}\right)$ and acetone $\left(10 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$ was added aqueous sodium metaperiodate ( 0.29 g ) in water ( 30 $\mathrm{cm}^{3}$ ). After 2 h TLC examination (system A) revealed that the starting material had been completely converted into a mixture of a major ( $R_{\mathrm{F}} 0.48$ ) and a minor ( $R_{\mathrm{F}} 0.36$ ) product. The solution was evaporated to dryness and the residue extracted with chloroform ( $3 \times 20 \mathrm{~cm}^{3}$ ). The extract was washed with water ( $3 \times 20 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to give a red solid ( 0.45 g ), m.p. $213^{\circ} \mathrm{C}$ (decomp.) presumably the $O$-formyl derivative (17a) (Found: $M^{+}, 400 . \mathrm{C}_{20} \mathrm{H}_{16} \mathrm{O}_{9}$ requires $M, 400$ ); $m / z 400\left(M^{+}\right), 382\left(M-\mathrm{H}_{2} \mathrm{O}\right), 270,269,254$, and $240(100 \%)$.
A solution of the product in methanol ( $15 \mathrm{~cm}^{3}$ ) and THF ( 50 $\mathrm{cm}^{3}$ ) at $0^{\circ} \mathrm{C}$ was mixed with $2 \%$ aqueous sodium hydroxide ( 25 $\mathrm{cm}^{3}$ ) containing sodium dithionite ( 0.174 g ) and left for 30 min in a nitrogen atmosphere. Air was passed through the solution for 1 h ; evaporation of the solution left a red solid which was treated with water and collected. The major component ( $R_{\mathrm{F}}$ 0.06) was identical on TLC examination in solvent systems (A) and (B) with the compound isolated under (a) above. It also had $m / z 372\left(M^{+}\right), 336\left(M-2 \mathrm{H}_{2} \mathrm{O}, 100 \%\right)$, 298, and 280.
(c) A solution of the aldehydoerythrose (21) ( 0.1 g ) in $\mathrm{m} / 2$
aqueous sodium hydroxide $\left(10 \mathrm{~cm}^{3}\right)$ with sodium dithionite $(0.1 \mathrm{~g})$ in $\mathrm{m} / 2$ aqueous sodium hydroxide $\left(2 \mathrm{~cm}^{3}\right)$ was left in a nitrogen atmosphere for 1 h . The solution was aerated for 1 h and the resulting purple solution added to $2 \mathrm{M}-\mathrm{HCl}\left(4 \mathrm{~cm}^{3}\right)$ and crushed ice ( 4 g ) with stirring. The red solid precipitate was collected and found to be identical (TLC examination in several solvent systems) to the compound obtained in ( $a$ ) and ( $b$ ) above.

5-Hydroxy-2,3,4a,9a-tetrahydroanthracene-1,4,9,10-tetraone (1b). -A solution of $1,4,5$-trihydroxyanthraquinone ( 6.0 g ) in aqueous sodium hydroxide $\left(2 \% ; 500 \mathrm{~cm}^{3}\right)$ was heated at $80^{\circ} \mathrm{C}$ for 1 h with sodium dithionite ( 12.0 g ); the reaction was kept under a nitrogen atmosphere throughout. The solution was allowed to cool, and then acidified with an excess of $5 \mathrm{M}-\mathrm{HCl}(100$ $\mathrm{cm}^{3}$ ). A yellow-brown precipitate of the title compound was collected, washed with water, and dried in vacuo $\left(\mathrm{P}_{2} \mathrm{O}_{5}\right)$. The product was recrystallised from pyridine.

Reaction of the Tetraone (1b) with 3-O-Benzyl-3-C-methyl-1,2-O-isopropylidene- $\alpha$-D-ribo-pentodialdofuranose using 1,5-Di-azabicyclo[4.3.0]non-5-ene (DBN) as the Base.-To a solution of 5-hydroxy-2,3,4a,9a,tetrahydroanthracene-1,4,9,10-tetraone (1b) ( 2.8 g ) in dry DMF $\left(50 \mathrm{~cm}^{3}\right)$ cooled to $0^{\circ} \mathrm{C}$, was added a solution of 3-O-benzyl-3-C-methyl-1,2- $O$-isopropylidene- $\alpha$-D-ribo-pentodialdofuranose ( 3.5 g ) in dry DMF ( $20 \mathrm{~cm}^{3}$ ) followed by DBN ( $3.5 \mathrm{~cm}^{3}$ ). After 45 min under nitrogen, TLC examination (system A) after aeration, revealed the complete disappearance of starting material and the presence of two products of equal intensity ( $R_{\mathrm{F}} 0.72$ ) and $R_{\mathrm{F}} 0.78$ ), a minor product ( $R_{\mathrm{F}} 0.82$ ), and some unchanged tetraone ( $R_{\mathrm{F}} 0.91$ ).
The brownish red reaction mixture was aerated for 90 min and the resultant purple solution poured into a mixture of crushed ice $(40 \mathrm{~g})$ and $2 \mathrm{M}-\mathrm{HCl}\left(40 \mathrm{~cm}^{3}\right)$ with rapid stirring. The bright red precipitate was best collected by extraction with chloroform ( $4 \times 50 \mathrm{~cm}^{3}$ ). The extract was washed with water $\left(2 \times 50 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$, and evaporated to provide a bright red solid ( 5.3 g ).
The foregoing solid was dissolved in toluene-ethyl acetate (4:1) and the solution applied to a silica gel column $(8 \times 60$ cm ) and eluted by the same solvent mixture. Products: compound $\mathrm{A}\left(R_{\mathrm{F}} 0.82,0.123 \mathrm{~g}\right)$, compound $\mathrm{B}\left(R_{\mathrm{F}} 0.78,1.1 \mathrm{~g}\right)$, and compound $\mathrm{C}\left(R_{\mathrm{F}} 0.72,1.2 \mathrm{~g}\right)$ were isolated.

Compound A. 5-Deoxy-3-O-benzyl-3-C-methyl-1,2-O-iso-propylidene-5-( $9^{\prime}, 10^{\prime}$-dihydro- $1^{\prime}, 4,5-$ trihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$ -anthryl)- $\alpha$-D-ribofuranose hemihydrate ( 9 g ) separated from toluene as red needles, m.p. $159^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 66.6 ; \mathrm{H}, 5.3 \%$; $M^{+}, 532 . \mathrm{C}_{30} \mathrm{H}_{28} \mathrm{O}_{9} \cdot \frac{1}{2} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 66.5 ; \mathrm{H}, 5.2 \% ; M, 532$ ); $m / z 532\left(M^{+}\right), 517\left(M-\mathrm{CH}_{3}\right), 325(100 \%), 270,269$, and 256 ; $\lambda_{\max }(\mathrm{EtOH}) 520,514 \mathrm{sh}, 494,481 \mathrm{sh}, 466 \mathrm{sh}$, and 390 sh ; $\delta-$ $\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.20\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.30$ and 1.45 $\left(2 \times 3 \mathrm{H}, \mathrm{CMe}_{2}\right), 2.75$ and $2.95(2 \times 1 \mathrm{H}, \mathrm{dd}, 5-\mathrm{H}), 4.30(1 \mathrm{H}$, dd, 4-H), 4.49 and $4.61\left(1 \times 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.53(1 \mathrm{H}, \mathrm{d}, 2-\mathrm{H})$, 7.20-7.23 ( $\left.5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.35\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), 7.40,7.74$ and $7.80\left(3 \times 1 \mathrm{H}, 6^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}, 8^{\prime}-\mathrm{H}\right), 12.20\left(2 \mathrm{H}, \mathrm{br}, 5^{\prime}-\mathrm{OH}\right.$ and $4^{\prime}-$ OH each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$, and $13.50\left(1 \mathrm{H}, \mathrm{s}, \mathrm{br}, 1^{\prime}-\mathrm{OH}\right)$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$.
Compound B. (5S)-3-O-Benzyl-3-C-methyl-1,2-O-isopropyl-idene-5-(9',10'-dihydro-1', $4^{\prime}, 5$-trihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$ -anthryl)- $\alpha$-D-ribofuranose ( 9 f ) separated from toluene as red needles, m.p. $>200^{\circ} \mathrm{C}$ (Found: C, $65.9 ; \mathrm{H}, 5.1 \% ; M^{+}, 548$. $\mathrm{C}_{30} \mathrm{H}_{28} \mathrm{O}_{10}$ requires C, $\left.65.7 ; \mathrm{H}, 5.15 \% ; M, 548\right) ; \mathrm{m} / \mathrm{z} 548\left(M^{+}\right)$, $530\left(M-\mathrm{H}_{2} \mathrm{O}\right), 490\left(M-\mathrm{Me}_{2} \mathrm{CO}\right), 472\left(M-\mathrm{Me}_{2} \mathrm{CO}-\right.$ $\mathrm{H}_{2} \mathrm{O}$ ), 286, 285 (5-hydroxyquinizarin- 2 -yl: $\mathrm{CHOH}, 100 \%$ ), 283, and 256; $\delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.20\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.30$ and $1.45\left(2 \times 3 \mathrm{H}, \mathrm{CMe}_{2}\right), 4.30\left(1 \mathrm{H}, \mathrm{d}, J_{4.5} 4.5 \mathrm{~Hz}, 4-\mathrm{H}\right), 4.50$ $\left(1 \mathrm{H}, \mathrm{d}, J_{2,1} 3.5 \mathrm{~Hz}, 2-\mathrm{H}\right), 4.35$ and $4.52\left(2 \times 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right)$, $4.65\left(1 \mathrm{H}, \mathrm{d}, J_{5.4} 4.5 \mathrm{~Hz}, 5-\mathrm{H}\right), 5.25\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 3.5 \mathrm{~Hz}, 1-\mathrm{H}\right)$, $5.85\left(1 \mathrm{H}, \mathrm{d}, 5-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.15-7.30\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.35$ $\left(1 \mathrm{H}, \mathrm{dd}, 7^{\prime}-\mathrm{H}\right), 7.50\left(1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), 7.75\left(1 \mathrm{H}, \mathrm{dd}, 8^{\prime}-\mathrm{H}\right), 7.90(1$
$\left.\mathrm{H}, \mathrm{t}, 6^{\prime}-\mathrm{H}\right), 12.10\left(1 \times 2 \mathrm{H}, \mathrm{br}, 5^{\prime}-\mathrm{OH}, 4^{\prime}-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$, and 13.45 ( $1 \mathrm{H}, \mathrm{br}, 1^{\prime}-\mathrm{OH}$ exch. $\mathrm{D}_{2} \mathrm{O}$ ).

Compound C. (5R)-3-O-Benzyl-3-C-methyl-1,2-O-isopropyl-idene-5-( $9^{\prime}, 10^{\prime}$-dihydro-1', $4^{\prime}, 5-$ trihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$ -
anthryl)- $\alpha$-D-ribofuranose (9e) separated from toluene as a hemihydrate, m.p. $>200^{\circ} \mathrm{C}$ (Found: C, 64.9; H, $5.1^{\circ} \% M^{+}$, 548; $\mathrm{C}_{30} \mathrm{H}_{28} \mathrm{O}_{10} \frac{-1}{2} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 64.65 ; \mathrm{H}, 5.25 \% ; M, 548$ ); $\lambda_{\text {max }}(\mathrm{EtOH}) 532,520 \mathrm{sh}, 496,482 \mathrm{sh}$, and 358sh; $m / z 548\left(M^{+}\right)$ $530\left(M-\mathrm{H}_{2} \mathrm{O}\right), 286,285(100 \%)$, 283 , and $256 ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right.$; $400 \mathrm{MHz}] 1.25\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 1.35$ and $1.45(2 \times 3 \mathrm{H}$, $\mathrm{CMe}_{2}$ ), $4.33\left(1 \mathrm{H}, \mathrm{d}, J_{4.5} 5.5 \mathrm{~Hz}, 4-\mathrm{H}\right), 4.45\left(1 \mathrm{H}, \mathrm{d}, J_{2,1} 3.5 \mathrm{~Hz}\right.$, 2-H), 4.39 and $4.52\left(2 \times 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.25(1 \mathrm{H}, \mathrm{t}$, collapses to doublet with $\left.\mathrm{D}_{2} \mathrm{O}, J_{5,4} 5.5 \mathrm{~Hz}, 5-\mathrm{H}\right), 5.69(1 \mathrm{H}, \mathrm{d}, \mathrm{br}, 5-\mathrm{OH}$, exch. $\mathrm{D}_{2} \mathrm{O}$ ), $7.08\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.40\left(1 \mathrm{H}, \mathrm{dd}, 7^{\prime}-\mathrm{H}\right), 7.54$ (1 $\left.\mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), 7.78\left(1 \mathrm{H}, \mathrm{dd}, 8^{\prime}-\mathrm{H}\right), 7.85\left(1 \mathrm{H}, \mathrm{t}, 6^{\prime}-\mathrm{H}\right), 12.10$ $\left(1 \times 2 \mathrm{H}, \mathrm{br}, 5^{\prime}-\mathrm{OH}, 4^{\prime}-\mathrm{OH}\right.$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$, and $13.80\left(1 \mathrm{H}, \mathrm{s}, \mathrm{br}, 1^{\prime}-\right.$ OH exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(5R)-3-O-Benzyl-3-C-methyl-5-( $9^{\prime}, 10^{\prime}$-dihydro- $1^{\prime}, 4^{\prime}, 5^{\prime}$-tri-hydroxy- $9^{\prime}, 10^{\prime}$-dioxo-2'-anthryl)- $\alpha$ - $\beta$-D-ribopyranose (15b).-A solution of the foregoing compound $\mathrm{C}(1.1 \mathrm{~g})$ in $70 \%$ aqueous acetic acid ( $140 \mathrm{~cm}^{3}$ ) was boiled under reflux for 3 h when TLC examination (system A) revealed that the starting material ( $\boldsymbol{R}_{\mathrm{F}}$ 0.72 ) had been replaced by a single product $R_{F} 0.32$ ). Evaporation of the solution gave a red solid. The ribopyranose ( 1.02 g ) crystallised from toluene as deep red needles which retained water, m.p. $203{ }^{\circ} \mathrm{C}$ (Found: C, 62.2; H, $4.6 \% ; M^{+}, 508$. $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{O}_{10} \cdot \frac{2}{3} \mathrm{H}_{2} \mathrm{O}$ requires C, $62.2 ; \mathrm{H}, 4.6 \% ; M, 508$ ); $m / z 508$ $\left(M^{+}\right), 490\left(M-H_{2} \mathrm{O}\right), 286,285,283$, and 256 ; $\lambda_{\max }(\mathrm{EtOH}) 532$, $519 \mathrm{sh}, 497,485 \mathrm{sh}, 469 \mathrm{sh}$, and $393 \mathrm{sh} ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right]$ $1.47\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.6\left(1 \mathrm{H}, \mathrm{d}, J_{4.5} 10 \mathrm{~Hz}, 4-\mathrm{H}\right), 3.81(1 \mathrm{H}$, dd, $2-\mathrm{H}), 4.55-4.6\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.1(1 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 1-\mathrm{H})$, $5.45(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}, 5-\mathrm{H}), 5.05,5.58,6.70(3 \times 1 \mathrm{H}$, br each exch. $\left.\mathrm{D}_{2} \mathrm{O}, 1-\mathrm{OH}, 2-\mathrm{OH}, 4-\mathrm{OH}\right), 7.35-7.50\left(7 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}, 3^{\prime}-\mathrm{H}\right.$, $\left.7^{\prime}-\mathrm{H}\right), 7.80-7.90\left(1 \times 2 \mathrm{H}, 6^{\prime}-\mathrm{H}, 8^{\prime}-\mathrm{H}\right), 12.10(1 \times 2 \mathrm{H}, \mathrm{br}$ $5^{\prime}-\mathrm{OH}, 4^{\prime}-\mathrm{OH}$ each exch. $\mathrm{D}_{2} \mathrm{O}$ ), and $13.45\left(1 \mathrm{H}, \mathrm{br}, 1^{\prime}-\mathrm{OH}\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$.
(4R)-2-O-Benzyl-2-C-methyl-3-O-formyl-4-(9',10'-dihydro$1^{\prime}, 4^{\prime}, 5^{\prime}$-trihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$-anthryl)- $\alpha(\beta)$-D-erythrotetrafuranose (18b).-To a solution of the foregoing ribopyranose derivative ( 1 g ) in acetic acid ( $120 \mathrm{~cm}^{3}$ ) at room temperature was added a solution of sodium metaperiodate ( 0.46 g ) in water ( $60 \mathrm{~cm}^{3}$ ) and acetone ( $40 \mathrm{~cm}^{3}$ ). After 2.5 h TLC examination (system A) revealed that the starting material ( $R_{F}$ 0.32 ) had been replaced by one major product ( $R_{\mathrm{F}} 0.74$ ) and a trace of a second compound ( $R_{F} 0.68$ ). The mixture was diluted with water ( $100 \mathrm{~cm}^{3}$ ) and extracted with chloroform ( $3 \times 50$ $\mathrm{cm}^{3}$ ). The combined extracts were washed with water ( $2 \times 50$ $\mathrm{cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to an orange-red solid. The erythro-tetrafuranose hemihydrate $(0.92 \mathrm{~g})$ separated from ethanol as red needles, m.p. $182^{\circ} \mathrm{C}$ (Found: C, 63.3; H, $4.5 \%$; $M^{+}, 506 . \mathrm{C}_{27} \mathrm{H}_{22} \mathrm{O}_{10} \frac{-1}{2} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 62.9 ; \mathrm{H}, 4.5 \% ; M, 506$ ); $m / z 506\left(M^{+}\right), 285,283 ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.34(1 \times 3$ $\left.\mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 4.58,4.63\left(1 \times 2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.25(1 \mathrm{H}, \mathrm{d}, 4-\mathrm{H})$, $5.40(1 \mathrm{H}, \mathrm{d}, 3-\mathrm{H}), 5.47\left(1 \mathrm{H}, \mathrm{d}, J_{1,1 \mathrm{OH}} 6 \mathrm{~Hz}, 1-\mathrm{H}\right), 6.80(1 \mathrm{H}, \mathrm{d}$, $J_{1-\mathrm{oH}, 1} 6 \mathrm{~Hz}, 1-\mathrm{OH}$, exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.25-7.50\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.68$ ( $\left.1 \mathrm{H}, \mathrm{s}, 3^{\prime}-\mathrm{H}\right), 7.78-7.86\left(1 \times 3 \mathrm{H}, 6^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}, 8^{\prime}-\mathrm{H}\right), 8.35(1 \mathrm{H}$, d, OCHO), and 12.0-12.5 ( $3 \mathrm{H}, \mathrm{br}, 5^{\prime}-\mathrm{OH}, 1^{\prime}-\mathrm{OH}, 4^{\prime}-\mathrm{OH}$, each exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(4R)-2-O-Benzyl-2-C-methyl-3-O-formyl-4-(1,2,3,4,4a,9,9a,-10-octahydro-5-hydroxy-1,4,9,10-tetraoxo-2-anthryl)- $\alpha(\beta)$ -D-erythro-tetrafuranose.-To a solution of the foregoing tetrafuranose ( 21 a ) ( 0.51 g ) in acetic acid ( $50 \mathrm{~cm}^{3}$ ) and methanol ( $15 \mathrm{~cm}^{3}$ ) activated zinc powder ( 1.5 g ) was added. The mixture was shaken at room temperature for 3 h and then filtered. The yellow filtrate was concentrated to $c a .20 \mathrm{~cm}^{3}$ and then
extracted with chloroform ( $3 \times 25 \mathrm{~cm}^{3}$ ). The combined extracts were washed with water and dilute hydrochloric acid, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to afford the leuco derivative as a yellow solid ( 0.5 g ), homogeneous on TLC examination ( $R_{F}$ 0.80 , system A).
(8R,9S,10R)-8-O-Benzyl-8-methyl-7,8,9,10-tetrahydro-4,6,8,-9,10,11-hexahydroxynaphthacene-5,12-dione (19h).-To a solution of the foregoing leuco derivative ( 0.5 g ) in DMF (8 $\mathrm{cm}^{3}$ ) at $0^{\circ} \mathrm{C}$ was added a solution of DBN $\left(0.2 \mathrm{~cm}^{3}\right)$ in DMF $\left(4 \mathrm{~cm}^{3}\right)$. The mixture was set aside in a nitrogen atmosphere for 40 min when TLC examination (system A) indicated that the starting material ( $R_{\mathrm{F}} 0.80$ ) had been replaced by two major products $\mathrm{A}\left(R_{\mathrm{F}} 0.28\right)$ and $\mathrm{B}\left(R_{\mathrm{F}} 0.22\right)$. Air was passed through the solution for 2 h and the resultant purple solution poured into a mixture of crushed ice $(50 \mathrm{~g})$ and $2 \mathrm{M}-\mathrm{HCl}\left(25 \mathrm{~cm}^{3}\right)$ to produce an orange-red precipitate which was collected, washed with water, and dried. The two products were separated on preparative TLC plates ( $20 \times 20 \mathrm{~cm}^{3}$ ) using toluene-ethyl acetate $(4: 1)$ as the eluant. The hexahydroxynaphthacene (compound A) ( 0.16 g ) formed an orange-red solid homogeneous on TLC; $m / z 462\left(M^{+}\right), 444\left(M-\mathrm{H}_{2} \mathrm{O}\right), 353$ $(100 \%), 336\left(M-\mathrm{H}_{2} \mathrm{O}-\mathrm{C}_{7} \mathrm{H}_{7} \mathrm{OH}\right) ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right]$ $1.55\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.8$ and $3.2(1 \times 2 \mathrm{H}, 9-\mathrm{OH}, 7-\mathrm{OH}$, each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 3.45$ and $3.5\left(2 \times 1 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.85$ and 5.1 $\left(2 \times 1 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 7-\mathrm{H}_{\mathrm{eq}}\right.$ and $\left.7-\mathrm{H}_{\mathrm{ax}}\right), 5.15(1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz}, 9-\mathrm{H}$, or $10-\mathrm{H}), 5.2(1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz}, 10-\mathrm{H}$ or $9-\mathrm{H}), 7.2-7.3(5 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{C}_{6} \mathrm{H}_{5}\right), 7.55,7.75$, and $7.9(3 \times 1 \mathrm{H}, 1-\mathrm{H}, 2-\mathrm{H}$, and $3-\mathrm{H}), 12.4$, 12.95 , and $13.55(3 \times 1 \mathrm{H}, 4-\mathrm{OH}, 5-\mathrm{OH}, 11-\mathrm{OH}$ each exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(7R/S)-8R,9S,10R-8-O-Benzyl-8-methyl-7,8,9,10-tetrahydro-4,6,7,8,9,10,11-heptahydroxynaphthacene-5,12-dione (19i).Compound $B\left(R_{F} 0.22\right)$ isolated in the foregoing reaction was obtained as an orange-red solid ( 0.17 g ) homogeneous on TLC; $m / z 478\left(M^{+}\right), 460\left(M-\mathrm{H}_{2} \mathrm{O}\right), 335\left(M-\mathrm{H}_{2} \mathrm{O}-\mathrm{C}_{7} \mathrm{H}_{7} \mathrm{OH}\right)$, 3i4, 296 (retro-Diels-Alder fragment), $91\left(100 \%, \mathrm{C}_{7} \mathrm{H}_{7}^{+}\right)$; $\delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.71\left(1 \times 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.65$ and 2.88 $\left(1 \times 2 \mathrm{H}, \mathrm{br}, 7-\mathrm{OH}, 9-\mathrm{OH}\right.$, each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.15-4.2(1 \times 2$ $\mathrm{H}, \mathrm{br}, J_{10.9} 8 \mathrm{~Hz}, 10-\mathrm{H}$ or $9-\mathrm{H}$ and $10-\mathrm{H}$, collapses to a doublet with $\left.\mathrm{D}_{2} \mathrm{O}\right), 4.65$ and $4.75\left(2 \times 1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.05$ $\left(1 \mathrm{H}, \mathrm{d}\right.$, collapses to a singlet with $\left.\mathrm{D}_{2} \mathrm{O}, 7-\mathrm{H}\right), 5.2\left(1 \mathrm{H}, \mathrm{d}, J_{9,10}\right.$ $8 \mathrm{~Hz}, 9-\mathrm{H}$ or $10-\mathrm{H}), 7.1-7.2\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.33,7.64$, and 7.85 $(3 \times 1 \mathrm{H}, 1-\mathrm{H}, 2-\mathrm{H}, 3-\mathrm{H}), 12.1,12.75$, and $13.85(3 \times 1 \mathrm{H}$, $4-\mathrm{OH}, 6-\mathrm{OH}$, and 11-OH, each exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(5S)-3-O-Benzyl-3-C-methyl-5-(9',10'-dihydro-1', $4^{\prime}, 5^{\prime}$-tri-hydroxy- $9^{\prime}, 10^{\prime}$-dioxo-2-anthryl)- $\alpha, \beta$-D-ribopyranose (14c).-A solution of (5S)-3-O-benzyl-3-C-methyl-1,2-O-isopropylidene-5-(9', $10^{\prime}$-dihydro- $1^{\prime}, 4^{\prime}, 5^{\prime}$-trihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $2^{\prime}$-anthryl)-$\alpha$-D-ribofuranose (9b) ( 1.1 g ) in $70 \%$ aqueous acetic acid ( 40 $\mathrm{cm}^{3}$ ) was boiled under reflux for 3.5 h when TLC examination (system A) indicated that the starting material had been completely replaced by a single new product ( $R_{\mathrm{F}} 0.34$ ). The solution was evaporated to dryness to give the title compound $(1.02 \mathrm{~g})$, m.p. $>200^{\circ} \mathrm{C}$ (Found: C, 63.8; H, $4.7 \% ; M^{+}, 508$. $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{O}_{10}$ requires C, 63.7; $\mathrm{H}, 4.7 \% ; M, 508$ ); $\mathrm{m} / \mathrm{z} 508\left(M^{+}\right)$, $490\left(M-\mathrm{H}_{2} \mathrm{O}\right), 286,285,284,283$, and $256 ; \delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} 400\right.$ $\mathrm{MHz} 1.5\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.51\left(1 \mathrm{H}, \mathrm{d}, J_{4.5} 4 \mathrm{~Hz}, 4-\mathrm{H}\right), 3.51(1$ $\mathrm{H}, \mathrm{dd}, 2-\mathrm{H}), 4.55-4.60\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.20(1 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 1-$ H), $5.25\left(1 \mathrm{H}, \mathrm{d}, J_{5.4} 4 \mathrm{~Hz}, 5-\mathrm{H}\right), 6.55,5.56$, and $5.01(3 \times 1 \mathrm{H}$, all d, 1-OH, 2-OH and 4-OH, each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.30-7.50(7 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{C}_{6} \mathrm{H}_{5}, 3^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}\right), 7.56-7.58\left(1 \times 2 \mathrm{H}, 6^{\prime}-\mathrm{H}, 8^{\prime}-\mathrm{H}\right), 12.201 \times 2$ $\mathrm{H}, \mathrm{br}, 5^{\prime}-\mathrm{OH}, 4^{\prime}-\mathrm{OH}$, each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right)$, and $13.5\left(1 \mathrm{H}, 1^{\prime}-\mathrm{OH}\right.$, exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(4S)-2-O-Benzyl-2-C-methyl-3-O-formyl-4-(9',10'-dihydro$1^{\prime}, 4^{\prime}, 5^{\prime}$-trihydroxy- $9^{\prime}, 10^{\prime}$-dioxo- $\mathbf{2}^{\prime}$-anthryl)- $\alpha, \beta$-D-erythro-tetra-
furanose ( $\mathbf{1 7 c}$ ).-To a solution of the ribopyranose (20b) (0.5 g) in acetic acid ( $60 \mathrm{~cm}^{3}$ ) at room temperature was added sodium metaperiodate ( 0.23 g in water $\left(30 \mathrm{~cm}^{3}\right)$ and acetone ( 20 $\mathrm{cm}^{3}$ ). The reaction was monitored by TLC examination system A), which revealed the complete disappearance of starting material ( $R_{\mathrm{F}} 0.34$ and the presence of a new single product ( $R_{\mathrm{F}}$ 0.82 ), after 2 h . Water ( $50 \mathrm{~cm}^{3}$ ) was added and the mixture extracted with chloroform ( $3 \times 25 \mathrm{~cm}^{3}$ ); the combined extracts were washed with water, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to give (the title compound ( 0.48 g ); $m / z 506\left(M^{+}\right), 285,284,283$, and 256; $\left.\delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right)\right] ; 1.35\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 4.51$ and $4.52\left(1 \times 2 \mathrm{H}, \mathrm{CH}_{2}-\mathrm{Ph}\right), 5.25$ and $5.40(2 \times 1 \mathrm{H}, \mathrm{d}, 2-\mathrm{H}$ and $4-\mathrm{H}), 5.5\left(1 \mathrm{H}, \mathrm{d} J_{1,1-\mathrm{OH}} 6 \mathrm{~Hz}, 1-\mathrm{H}\right), 6.80\left(1 \mathrm{H}, \mathrm{d}, J_{1-\mathrm{OH}, 1} 6\right.$ $\mathrm{Hz}, 1-\mathrm{OH}$ exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 7.35-7.45\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.70(1 \mathrm{H}, \mathrm{s}$, $\left.3^{\prime}-\mathrm{H}\right), 7.85-7.90\left(1 \times 3 \mathrm{H}, 6^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}\right.$, and $\left.8^{\prime}-\mathrm{H}\right), 8.40(1 \mathrm{H}, \mathrm{d}$, OCHO), 12.0-13.0 ( $3 \mathrm{H}, \mathrm{br}, 5^{\prime}-\mathrm{OH}, 1^{\prime}-\mathrm{OH}$, and $4^{\prime}-\mathrm{OH}$ each exch. $\mathrm{D}_{2} \mathrm{O}$ ).
(8R,9S,10S)-8-O-Benzyl-8-C-methyl-7,8,9,10-tetrahydro-4,6,8,9,10,11-hexahydroxynaphthacene-5,12-dione (19j).-To a solution of the foregoing erythro-tetrafuranose ( 0.255 g ) in acetic acid ( $30 \mathrm{~cm}^{3}$ ) was added methanol ( $10 \mathrm{~cm}^{3}$ ) and zinc powder ( 0.5 g ) and the mixture stirred at room temperature for 1.5 h . After this it was filtered and the filtrate was evaporated to $c a .15 \mathrm{~cm}^{3}$ and extracted with chloroform ( $3 \times 20 \mathrm{~cm}^{3}$ ). The combined extracts were washed with water and dilute hydrochloric acid, dried $\left(\mathrm{MgSO}_{4}\right)$, and evaporated to give a yellow solid $(0.24 \mathrm{~g})$ of presumably (4S)2-O-benzyl-2-C-methyl-3-O-formyl-4-(1,2,3,4,4a,9,9a,10-octahydro-5-hydroxy-1,4,9,10-tetraoxo-2-anthryl)-D-erythro-tetrafuranose. The solid was dissolved in dry DMF $\left(5 \mathrm{~cm}^{3}\right)$ cooled to $0^{\circ} \mathrm{C}$ and then treated with DBN $\left(0.1 \mathrm{~cm}^{3}\right)$. The solution was kept in a nitrogen atmosphere for 25 min when TLC examination (system A) indicated that the starting material ( $R_{\mathrm{F}} 0.84$ ) had been replaced by two products $R_{\mathrm{F}} 0.36$ ) and ( $R_{\mathrm{F}} 0.30$ ). The yellow-brown solution was aerated for 1.5 h and the resulting purple solution poured into a mixture of $2 \mathrm{~m}-\mathrm{HCl}\left(10 \mathrm{~cm}^{3}\right)$ and crushed ice $(25 \mathrm{~g})$ to produce a red precipitate which was collected, washed with water, and dried. The two compounds were separated on preparative TLC plates ( $20 \times 20 \mathrm{~cm}$ ) using toluene-ethyl acetate (4:1) as the eluant. The hexahydroxynaphthacene ( $\mathrm{R}_{\mathrm{F}}$ 0.36 ) was obtained as a red solid ( 0.046 g ), homogeneous on TLC examination in systems A and B, $m / z 462\left(M^{+}\right), 444$ ( $M-\mathrm{H}_{2} \mathrm{O}$ ), 354, and $336\left(100 \%, M-\mathrm{H}_{2} \mathrm{O}-\mathrm{C}_{7} \mathrm{H}_{7} \mathrm{OH}\right)$; $\delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.51\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.54$ and 3.0 $\left(2 \times 1 \mathrm{H}, \mathrm{dd}, 9-\mathrm{OH}\right.$, and 7-OH each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 3.60(1 \times 2 \mathrm{H}$, d, $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 4.50$ and $4.52\left(1 \times 2 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 7-\mathrm{H}_{\mathrm{eq}}, 7-\mathrm{H}_{\mathrm{ax}}\right)$, $5.2(1 \mathrm{H}, \mathrm{d}, J 3 \mathrm{~Hz}, 9-\mathrm{H}), 5.25(1 \mathrm{H}, \mathrm{br}, 10-\mathrm{H}), 7.15-7.25(5 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.35,7.7$, and $7.9(1 \times 3 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}, 2-\mathrm{H}$, and $3-\mathrm{H})$, $12.20,12.65$, and $13.95(1 \times 3 \mathrm{H}, 4-\mathrm{OH}, 6-\mathrm{OH}, 11-\mathrm{OH}$, each exch. $\mathrm{D}_{2} \mathrm{O}$ ).
$(19 \mathrm{k})$.-The heptahydroxynaphthacene ( 0.05 g ) was obtained in the foregoing experiment as a red solid homogeneous on TLC examination with $R_{\mathrm{F}} 0.30 ; m / z 478\left(M^{+}\right), 460(M-$ $\mathrm{H}_{2} \mathrm{O}$ ), 354, $336\left(100 \%, M-\mathrm{H}_{2} \mathrm{O}-\mathrm{C}_{7} \mathrm{H}_{7} \mathrm{OH}\right)$, 314, 296; $\delta\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO} ; 400 \mathrm{MHz}\right] 1.43\left(1 \times 3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 2.53,3.15$, and $3.65\left(3 \times 1 \mathrm{H}, 7-\mathrm{OH}, 9-\mathrm{OH}, 10-\mathrm{OH}\right.$ each exch. $\left.\mathrm{D}_{2} \mathrm{O}\right), 3.45$ and $3.50\left(2 \times 1 \mathrm{H}, \mathrm{d}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.05(1 \mathrm{H}, \mathrm{dd}, 9-\mathrm{H}$ or $10-\mathrm{H}$ collapses to a d with $\left.\mathrm{D}_{2} \mathrm{O}\right), 5.35\left(1 \mathrm{H}, \mathrm{d}, J_{7,7 . \mathrm{OH}} 4 \mathrm{~Hz}, 7-\mathrm{H}\right.$, collapses to a s with $\mathrm{D}_{2} \mathrm{O}$ ), $7.2-7.3\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 7.35,7.70$ and 7.90 $(3 \times 1 \mathrm{H}, 1-\mathrm{H}, 2-\mathrm{H}$ and $3-\mathrm{H}), 12.1,12.85$ and $13.6(3 \times 1 \mathrm{H}$, $4-\mathrm{OH}, 6-\mathrm{OH}$ and $11-\mathrm{OH}$; each exch. $\mathrm{D}_{2} \mathrm{O}$ ).

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[^0]:    $\dagger$ For convenience, throughout the Discussion, leucoquinizarin (1a) and quinizarin-2-yl are used to refer to 2,3,4a,9a-tetrahydroanthracene-1,4,9,10-tetraone (1a) and 9,10-dihydro-9,10-dioxo-2-anthryl respectively. The abbreviated names are also used in the detailed spectroscopic results recorded in the Experimental section.

